

# Conjugate Acid-Base Pairs

Molecules or ions related by the loss/gain of **one**  $\text{H}^+$ .

| Conjugate Acid           |                       | Conjugate Base            |
|--------------------------|-----------------------|---------------------------|
| $\text{H}_3\text{O}^+$   |                       | $\text{H}_2\text{O}$      |
| $\text{CH}_3\text{COOH}$ | → donate $\text{H}^+$ | $\text{CH}_3\text{COO}^-$ |
| $\text{NH}_4^+$          | ← accept $\text{H}^+$ | $\text{NH}_3$             |
| $\text{H}_2\text{SO}_4$  |                       | $\text{HSO}_4^-$          |
| $\text{HSO}_4^-$         |                       | $\text{SO}_4^{2-}$        |
| $\text{HCl}$             |                       | $\text{Cl}^-$             |

$\text{NH}_4^+$  and  $\text{NH}_2^-$  are **not** conjugate  
(conversion requires 2  $\text{H}^+$ )

# Answers For Conjugate Acid Base Pairs

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