

Ternary compounds have _____ (three) _____ (three or) elements. They are usually always composed of a metal/oid _____ (metals) anhydride and acid or polyhydride _____ (anhydride) anion. The only positively charged polyhydride ion is _____ (ammonium ion). To write the formula of a ternary compound is no different than to write the formula of a binary compound with one exception. If a subscript is necessary for the polyhydride ion in order to _____ (balance) charge, you must place the polyhydride ion in _____ (parentheses).

As each ion, write the formula of the ionic compound consisting of the anion/cation on the left of the dashed line repeating an anion/cation the last.

	OH^-	SO_4^{2-}	NO_3^-	CO_3^{2-}	PO_4^{3-}
K^+	_____	_____	_____	_____	_____
Na^{2+}	_____	_____	_____	_____	_____
NH_4^+	_____	_____	_____	_____	_____
Al^{3+}	_____	_____	_____	_____	_____
Ca^{2+}	_____	_____	_____	_____	_____
Fe^{2+}	_____	_____	_____	_____	_____

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