

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	very	very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	A	nitric acid
$Mg(OH)_2$	B	magnesium hydroxide
H_3PO_4	A	phosphoric acid
H_2CO_3	A	carbonic acid
HF	A	hydrofluoric acid
$Fe(OH)_3$	B	iron (iii) hydroxide

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Acids And Bases Answer Key

Bryan M. Ham,Aihui MaHam

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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 - The Rise of Digital Reading Acids And Bases Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Acids And Bases Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acids And Bases Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acids And Bases Answer Key
 - Personalized Recommendations
 - Acids And Bases Answer Key User Reviews and Ratings
 - Acids And Bases Answer Key and Bestseller Lists
5. Accessing Acids And Bases Answer Key Free and Paid eBooks

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- Acids And Bases Answer Key eBook Subscription Services
- Acids And Bases Answer Key Budget-Friendly Options
- 6. Navigating Acids And Bases Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Acids And Bases Answer Key Compatibility with Devices
 - Acids And Bases Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acids And Bases Answer Key
 - Highlighting and Note-Taking Acids And Bases Answer Key
 - Interactive Elements Acids And Bases Answer Key
- 8. Staying Engaged with Acids And Bases Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acids And Bases Answer Key
- 9. Balancing eBooks and Physical Books Acids And Bases Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acids And Bases Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Acids And Bases Answer Key
 - Setting Reading Goals Acids And Bases Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Acids And Bases Answer Key
 - Fact-Checking eBook Content of Acids And Bases Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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