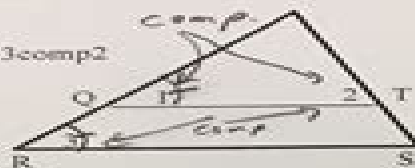


Name: _____

Key

1. Given: 1comp2, 3comp2

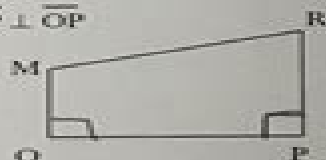
Prove: $\overline{QT} \parallel \overline{RS}$



- | | |
|---|--|
| <p>① $\angle 1 \text{ comp } \angle 2$
 $\angle 2 \text{ comp } \angle 3$</p> <p>② $\angle 1 \cong \angle 3$</p> <p>③ $\overline{QT} \parallel \overline{RS}$</p> | <p>① Given</p> <p>② Comp to Same $\rightarrow \cong$</p> <p>③ $\cong$ Corr. $\angle$'s $\rightarrow \parallel$</p> <p style="text-align: center;">QED</p> |
|---|--|

2. Given: $\angle MOP$ is RL $\overline{RP} \perp \overline{OP}$

Prove: $\overline{MO} \parallel \overline{RP}$

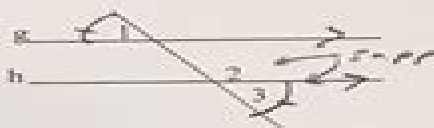


- | | |
|--|---|
| <p>① $\angle MOP$ is RL $\angle$</p> <p>$\overline{RP} \perp \overline{OP}$</p> <p>② $\overline{MO} \perp \overline{OP}$</p> <p>③ $\overline{MO} \parallel \overline{RP}$</p> | <p>① Given</p> <p>② RL $\angle \rightarrow \perp$</p> <p>③ PTT ($\perp$ to Same $\rightarrow \parallel$)</p> |
|--|---|

Per: _____

3. Given: $g \parallel h$

Prove: 1 supp 2



- | | |
|---|--|
| <p>① $g \parallel h$</p> <p>② $\angle 1 \cong \angle 3$</p> <p>③ $\angle 2 \text{ supp } \angle 3$</p> <p>④ $\angle 1 \cong \angle 3$</p> <p>⑤ $\angle 1 + \angle 2 = 180$</p> <p>⑥ $m\angle 2 + m\angle 1 = 180$</p> <p>⑦ $\angle 1 \text{ supp } \angle 2$</p> | <p>① Given</p> <p>② $\parallel \rightarrow \cong$ Alt. Ext. $\angle$'s</p> <p>③ LP $\rightarrow$ Supp</p> <p>④ $\cong \rightarrow =$</p> <p>⑤ Supp $\rightarrow 180$</p> <p>⑥ Subst. Prop. =</p> <p>⑦ $180 \rightarrow$ Supp</p> <p style="text-align: center;">QED</p> |
|---|--|

4. Given: $\angle C$ supp $\angle D$

Prove: $\angle A$ supp $\angle B$



- | | |
|---|---|
| <p>① $\angle C$ supp $\angle D$</p> <p>② $\overline{AD} \parallel \overline{BC}$</p> <p>③ $\angle A$ supp $\angle B$</p> | <p>① Given</p> <p>② S.S. Int + Supp $\rightarrow \parallel$</p> <p>③ $\parallel \rightarrow$ SS Int. Supp</p> <p style="text-align: center;">QED</p> |
|---|---|

14 Geometry Answers

Zhe Xu



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generated molecular equation chemical equation in which all reactants and products are represented as neutral substances

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whichever reactant gives the lesser amount of product is the limiting reagent what we need to do is determine an amount of one product either moles or mass assuming all of each reactant reacts

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