

Chapter 11.1

Describing Chemical Reactions

A. Introduction to Chemical Equations

4. What are the different symbols used in chemical equations?

Symbol	Explanation
+	Separates two reactants or two products
→	"yields", separates reactants from products
↔	Used in place of → for reversible reactions
(s), (l), (g)	Designates a reactant or product in the solid state, liquid state and gaseous state; placed after the formula
(aq)	Designates an aqueous solution; the substance is dissolved in water; placed after the formula
Heat →	Indicates that heat is supplied to the reaction
Pt→	A formula written above or below the yield sign indicates its use as a catalyst

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