

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the chemical equation
- 2. Explain how to interpret the equation
- 3. List the steps for writing a chemical equation

Vocabulary

- 1. chemical equation
- 2. chemical equation
- 3. reactant
- 4. coefficient
- 5. balanced equation

Part A Completion

Write the coefficient for each in the chemical equation, and balance the chemical equation. Write the name of each substance in the equation.

_____ a chemical equation is always represented by a symbol _____
 The chemical formula for a chemical compound is _____ The mass
 number for an element is represented by the _____ An element with
 the same chemical formula is _____ a chemical equation is a reaction
 representing a chemical reaction. The first _____ is the reactant and
 products are the substances that are formed. The _____ is the product. The
 equation is balanced when the number of atoms of each element is the same on
 both sides of the equation. The equation is balanced by changing the _____
 the chemical equation.

Chemical equations are written with the chemical formula for each
 substance. The chemical formula is _____ that is a symbol _____ and the
 _____ of the substance. Chemical equations are balanced _____ of _____
 mass is conserved. A chemical equation is balanced when the number of atoms of each
 element is the same on both sides of the equation.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
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18. _____
19. _____
20. _____

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Kevin T. Patton, Gary A. Thibodeau



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web nov 9 2009 if the conditions then really demand extra cooling then the stat will fully open the bypass system will close off by the disc shaped valve and the entire cooling system will consist of the block head outlet hose under the driver s seat radiator lower radiator tank thermostat expansion aux tank heaters et al until things cool