

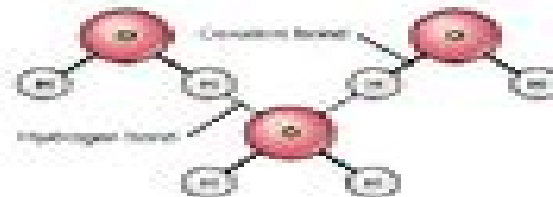
Unit 1 study guide

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Class	Unit 1 : Chemistry of Life
Reviewed	☐

Topic 1.1 : Structure of Water and Hydrogen Bonding

1. Structure /Polarity

- 2 elements : Oxygen and Hydrogen (1:2 ratio)
- Oxygen is more electronegative than hydrogen , resulting in an unequal sharing of electrons between oxygen and hydrogen. This is called Polarity, where there are differences in atoms electronegativity , a water molecule has polarity
- The bond is a covalent bond between the two elements (they share electrons)
- The bond between other water molecules are hydrogen bond (a weak bond between the negative and positive regions of two separate molecules).



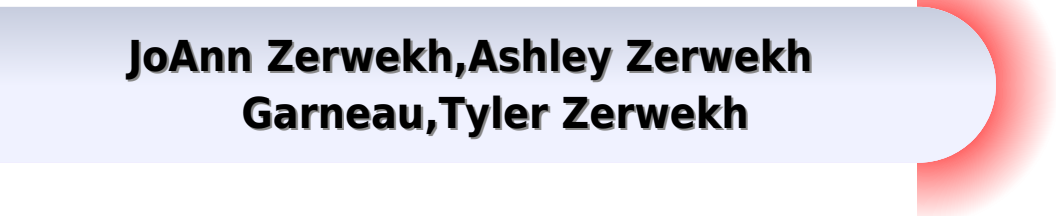
2. Properties of water

LIVING THINGS DEPEND ON WATER'S PROPERTIES

- When two of the same molecules form hydrogen bonds with each other this is called cohesion (water and water) .
- When two different molecules form hydrogen bonds with each other this is called adhesion (amino acid and water)

1study Guide Answers

**JoAnn Zerwekh, Ashley Zerwekh
Garneau, Tyler Zerwekh**



1study Guide Answers:

Study Guide for Nursing Research - E-Book Geri LoBiondo-Wood, Judith Haber, Carey Berry, 2021-07-14 UPDATED
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The Comprehensive Review Guide for Health Information Carla Tyson-Howard, Shirlyn C. Thomas, 2009-10-07 Perfect for preparing for the RHIA and RHIT exams the Comprehensive Review Guide for Health Information serves as a personal tutor for HIM students to review the major learning competencies that will be tested on the exams the workbook section divided by knowledge clusters contains short answer fill in the blank and matching questions to assess the students understanding of the competency after using the comprehensive review book section and listening to audio recordings At the end of every knowledge cluster a multiple choice test simulates the questions on the national RHIA and RHIT

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Illustrated Study

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