

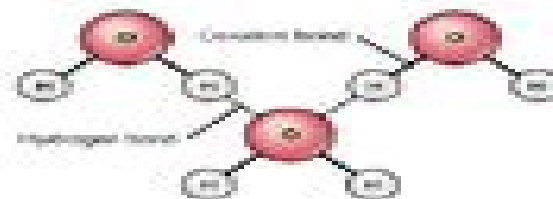
Unit 1 study guide

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Class	Unit 1 : Chemistry of Life
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Topic 1.1 : Structure of Water and Hydrogen Bonding

1. Structure /Polarity

- 2 elements : Oxygen and Hydrogen (1:2 ratio)
- Oxygen is more electronegative than hydrogen , resulting in an unequal sharing of electrons between oxygen and hydrogen. This is called Polarity, where there are differences in atoms electronegativity , a water molecule has polarity
- The bond is a covalent bond between the two elements (they share electrons)
- The bond between other water molecules are hydrogen bond (a weak bond between the negative and positive regions of two separate molecules).



2. Properties of water

LIVING THINGS DEPEND ON WATER'S PROPERTIES

- When two of the same molecules form hydrogen bonds with each other this is called cohesion (water and water) .
- When two different molecules form hydrogen bonds with each other this is called adhesion (amino acid and water)

1study Guide And Intervention

Kim Cooper, Kelly Gosnell



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Table of Contents 1study Guide And Intervention

1. Understanding the eBook 1study Guide And Intervention
 - The Rise of Digital Reading 1study Guide And Intervention
 - Advantages of eBooks Over Traditional Books
2. Identifying 1study Guide And Intervention
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 1study Guide And Intervention
 - User-Friendly Interface
4. Exploring eBook Recommendations from 1study Guide And Intervention
 - Personalized Recommendations
 - 1study Guide And Intervention User Reviews and Ratings
 - 1study Guide And Intervention and Bestseller Lists
5. Accessing 1study Guide And Intervention Free and Paid eBooks
 - 1study Guide And Intervention Public Domain eBooks
 - 1study Guide And Intervention eBook Subscription Services
 - 1study Guide And Intervention Budget-Friendly Options
6. Navigating 1study Guide And Intervention eBook Formats

- ePub, PDF, MOBI, and More
- 1study Guide And Intervention Compatibility with Devices
- 1study Guide And Intervention Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 1study Guide And Intervention
 - Highlighting and Note-Taking 1study Guide And Intervention
 - Interactive Elements 1study Guide And Intervention
- 8. Staying Engaged with 1study Guide And Intervention
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 1study Guide And Intervention
- 9. Balancing eBooks and Physical Books 1study Guide And Intervention
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 1study Guide And Intervention
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 1study Guide And Intervention
 - Setting Reading Goals 1study Guide And Intervention
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 1study Guide And Intervention
 - Fact-Checking eBook Content of 1study Guide And Intervention
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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