

2.3 Carbon Compounds

Lesson Summary

The Chemistry of Carbon Organic chemistry is the study of compounds with bonds between _____ atoms. Carbon atoms have _____ valence electrons, allowing them to form strong _____ bonds with many other elements, including hydrogen, oxygen, phosphorus, sulfur, and nitrogen. Living organisms are made up of molecules made of carbon and these other elements.

- ▶ One carbon atom can bond to another to form _____ and rings.
- ▶ Carbon can form _____ of different large and complex structures.

Macromolecules Many of the carbon molecules in living things are so large they are called _____. Macromolecules form by polymerization, in which smaller units called _____ join together to form _____. Biochemists sort the macromolecules in living things into groups based on their chemical composition.

- ▶ _____ (starches and sugars) are composed of carbon, hydrogen, and oxygen. Carbohydrates are the main _____ source for living things. Plants and some animals also use carbohydrates for structural purposes. Molecules with _____ sugar monomer are **monosaccharides**. A disaccharide is made of _____ monosaccharides.
- ▶ _____ (fats, oils, and waxes) are made mostly of carbon and hydrogen atoms. Lipids can be used to _____ energy and form parts of biological membranes and waterproof coverings. Steroids manufactured by the body are lipids as well. **LIPIDS ARE NOT POLYMERS!**
- ▶ _____ contain hydrogen, oxygen, nitrogen, carbon, and phosphorus. They are made by a monomer called _____. A nucleotide has three parts: a 5-carbon sugar, a phosphate ($-\text{PO}_4$) group, and a nitrogenous base. Nucleic acids store and transmit hereditary (_____) information. There are two kinds of nucleic acids: _____ (deoxyribonucleic acid) and _____ (ribonucleic acid).
- ▶ _____ are made up of nitrogen, carbon, hydrogen, and oxygen. Proteins are made by a monomer called _____. An amino acid molecule has an amino group ($-\text{NH}_2$) on one end and a carboxyl group ($-\text{COOH}$) on the other end. Proteins control the rate of _____, regulate cell processes, form cellular structures, carry substances into or out of cells, and help fight disease.
 - More than _____ different amino acids are found in nature. Any amino acid can bond with any other.
 - Covalent bonds called _____ bonds link amino acids together to form a polypeptide.
 - Amino acids are assembled into polypeptide chains according to instructions coded in DNA.

2 3 Carbon Compounds Answer Key

Victor von Richter



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