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## The Female Reproductive System – Day 3

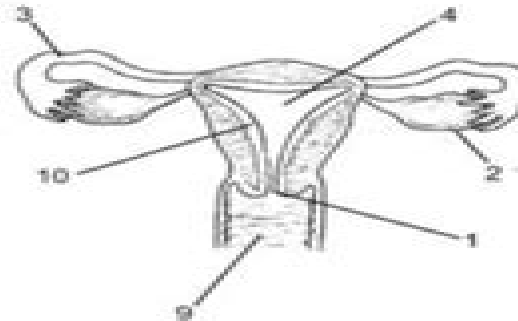
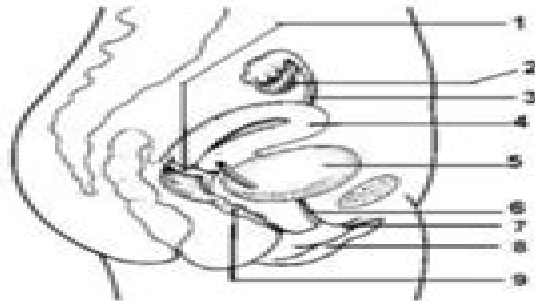
**Objective:** Students will be able to identify the parts and functions of the female reproductive system.

**QUESTION:** What part(s) of the female anatomy is (are) most susceptible to sexually transmitted diseases?

Student answers will vary. Acceptable hypotheses would focus on the cervix, vagina, and/or labia.

### The Female Reproductive System – Anatomy

Use the word bank below to identify the parts of the female reproductive system.



clitoris    ovary    labia (majora & minora)    uterus    cervix    urethra    vagina  
endometrium    fallopian tube (oviduct)    bladder

1. cervix
2. ovary
3. fallopian tube (oviduct)
4. uterus
5. bladder\* (not part of the reproductive system)

6. urethra\* (not part of the reproductive system)
7. clitoris
8. labia (majora & minora)
9. vagina
10. endometrium

### Check for Understanding:

1. Which 2 parts shown in the diagram are not part of the female reproductive system?

Bladder & urethra

2. Which part(s) is (are) on the outside of the female's body?

# Anatomy Of The Female Reproductive System Answer Key

**Ronald L. Eisenberg**



## **Anatomy Of The Female Reproductive System Answer Key:**

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web state one reason in terms of energy to support your answer answer endothermic the products have more energy than the reactants b on the diagram provided in your answer booklet draw a dashed line to indicate a potential energy curve for the reaction if a catalyst is added 46 given the potential energy diagram for a reaction

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web when the reactants have more potential energy than the products have and the extra energy is released to the surroundings it is called activation energy the height of the hill on a potential energy diagram is a measure of the endothermic the reaction above can be categorized as

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web this is like a one dimensional system whose mechanical energy  $e$  is a constant and whose potential energy with respect to zero energy at zero displacement from the spring  $s$  unstretched length  $x = 0$  is  $u = \frac{1}{2} kx^2$  figure 8 5 2 8 5 2 a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the  $pe$  of the reactants is 430kj and the  $pe$  of the products is 190kj what is the  $\Delta h$  2 if the  $pe$  of the reactants is 430kj and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web answer see below explanation the rate of a chemical reaction  $a + b \rightarrow c + d$  is given in first approximation by  $v = k \frac{[a]^m [b]^n}{[c]^p [d]^q}$  where  $k$  is the rate constant  $e_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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web key points potential energy is energy that has the potential to become another form of energy an object's potential energy depends on its physical properties and position in a system potential energy comes in many forms such as gravitational potential energy due to an object's mass and position in a gravitational field

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