

CHAPTER 7 REVIEW

Chemical Formulas and Chemical Compounds**SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. _____ In a Stock system name such as iron(III) sulfate, the Roman numeral tells us
 - (a) how many atoms of Fe are in one formula unit.
 - (b) how many sulfate ions can be attached to the iron atom.
 - (c) the charge on each Fe ion.
 - (d) the total positive charge of the formula unit.
2. _____ Changing a subscript in a correctly written chemical formula
 - (a) changes the number of moles represented by the formula.
 - (b) changes the charges on the other ions in the compound.
 - (c) changes the formula so that it no longer represents the compound it previously represented.
 - (d) has no effect on the formula.
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
 - _____ a. How many elements make up this compound?
 - _____ b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
 - _____ c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 - _____ d. How many atoms are present in a sample of 2.0×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
 - _____ a. $Ca(HCO_3)_2$
 - _____ b. $C_{12}H_{22}O_{11}$
 - _____ c. $Fe(ClO_2)_3$
 - _____ d. $Fe(ClO_3)_2$
5. _____
 - _____ a. What is the formula for the compound dinitrogen pentoxide?
 - _____ b. What is the Stock system name for the compound FeO ?
 - _____ c. What is the formula for sulfurous acid?
 - _____ d. What is the name for the acid H_3PO_4 ?

Answers Review Chemical Formulas And Chemical Compounds

Patrick Vollmar



Answers Review Chemical Formulas And Chemical Compounds:

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Table of Contents Answers Review Chemical Formulas And Chemical Compounds

1. Understanding the eBook Answers Review Chemical Formulas And Chemical Compounds
 - The Rise of Digital Reading Answers Review Chemical Formulas And Chemical Compounds
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Review Chemical Formulas And Chemical Compounds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Review Chemical Formulas And Chemical Compounds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Review Chemical Formulas And Chemical Compounds
 - Personalized Recommendations

- Answers Review Chemical Formulas And Chemical Compounds User Reviews and Ratings
- Answers Review Chemical Formulas And Chemical Compounds and Bestseller Lists
- 5. Accessing Answers Review Chemical Formulas And Chemical Compounds Free and Paid eBooks
 - Answers Review Chemical Formulas And Chemical Compounds Public Domain eBooks
 - Answers Review Chemical Formulas And Chemical Compounds eBook Subscription Services
 - Answers Review Chemical Formulas And Chemical Compounds Budget-Friendly Options
- 6. Navigating Answers Review Chemical Formulas And Chemical Compounds eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers Review Chemical Formulas And Chemical Compounds Compatibility with Devices
 - Answers Review Chemical Formulas And Chemical Compounds Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers Review Chemical Formulas And Chemical Compounds
 - Highlighting and Note-Taking Answers Review Chemical Formulas And Chemical Compounds
 - Interactive Elements Answers Review Chemical Formulas And Chemical Compounds
- 8. Staying Engaged with Answers Review Chemical Formulas And Chemical Compounds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Review Chemical Formulas And Chemical Compounds
- 9. Balancing eBooks and Physical Books Answers Review Chemical Formulas And Chemical Compounds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Review Chemical Formulas And Chemical Compounds
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers Review Chemical Formulas And Chemical Compounds
 - Setting Reading Goals Answers Review Chemical Formulas And Chemical Compounds
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers Review Chemical Formulas And Chemical Compounds
 - Fact-Checking eBook Content of Answers Review Chemical Formulas And Chemical Compounds

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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