

Net Ionic Equations Worksheet

Write balanced chemical equations for dissolving the following compounds in water:

1. Solid sodium chloride



2. Solid barium nitrate



3. Solid magnesium hydroxide



4. Gaseous hydrochloric acid



5. Solid calcium phosphate



6. Solid iron (III) sulfide



7. Gaseous carbon dioxide (*Hint: Carbon dioxide will form carbonic acid first then decomposes into ions*)



Write balanced chemical equations and then net ionic equations for the following reactions:

8. Aqueous sodium sulfate and aqueous barium chloride



9. Aqueous lithium bromide and aqueous silver nitrate



10. Aqueous lead II acetate and aqueous sodium carbonate



Answers For Pogil Net Ionic Reactions

KJ Lindholm-Leary



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