Write a balanced chemical equation to represent each of the following chemical reactions.

- 1. iron + sulfur iron(II) sulfide
- 2. zinc + copper(II) sulfate → zinc sulfate + copper
- 3. silver nitrate + sodium bromide → sodium nitrate + silver bromide
- potassium chlorate (heated) potassium chloride + oxygen
- water (electricity) → hydrogen + oxygen
- mercury(II) oxide (heated) → mercury + oxygen
- potassium iodide + lead(II) nitrate → lead(II) iodide + potassium nitrate
- 8. aluminum + oxygen → aluminum oxide
- 9. magnesium chloride + ammonium nitrate -- magnesium nitrate + ammonium chloride
- 10. iron(III) chloride + ammonium hydroxide → iron(III) hydroxide + ammonium chloride
- sodium peroxide + water → sodium hydroxide + oxygen
- iron(III) oxide + carbon → iron + carbon monoxide
- 13. iron + water → hydrogen + iron(III) oxide
- iron(III) chloride + potassium hydroxide → potassium chloride + iron(III) hydroxide
- 15. aluminum + sulfuric acid → aluminum sulfate + hydrogen
- 16. sodium carbonate + calcium hydroxide → sodium hydroxide + calcium carbonate

Answer Key Weston Walch Chemistry

Diane Helder, Allen Lampert

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Chemistry Thomas Kardos, 2003 This book provides a cornerstone for understanding atomic structure chemical bonding chemical reactions the periodic table and more It contains teacher demos and lab activities that stimulate scientific inquiry checked for safety and designed for easy inexpensive use

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