



NAME: Abhishek H. Nair Div: A Roll No. 1

SCIENCE Paper – 2*(Two hours)**You will not be allowed to write during the first 15 minutes.**This time is to be spent in reading the Question Paper.**The time given at the head of this paper is the time allowed for writing the answers.**Section I is compulsory. Attempt any four questions from Section II.**The intended marks for questions or parts of questions are given in brackets []***SECTION I (40 Marks)***(Attempt ALL questions)***Question 1****[15]**

Choose the correct answer from the options given below:

(Do not copy the question; write the correct answers only.)

- Whenever the gases react under similar conditions of temperature and pressure, they do so in volumes which bear a simple ratio of each other and the products if gaseous. The above law was stated by
 a) Boyle b) Charles c) Gay Lussac d) Avogadro
- A strip of copper is placed in four different colourless salt solutions. After electrolysis, one of the salt solutions becomes blue in colour. Identify the salt solution which turns blue from the given salt solutions:
 a) K_2SO_4 b) $CuSO_4$ c) $AgSO_4$ d) $CaSO_4$
- Potassium is the element below sodium in the long form of the periodic table. The electronegativity of potassium will be
 a) Lower than sodium b) Higher than sodium c) Same as sodium d) None of these
- The salt solution which does not react with ammonium hydroxide is :
 a) Calcium Nitrate b) Zinc Nitrate c) Lead Nitrate d) Copper Nitrate
- Assertion :** Sodium is the most reactive metal among group 1 elements.
Reason : In case of metals, greater the tendency to lose electrons, greater is the reactivity.
 a) A and R both are correct statements and R is the correct explanation of the A
 b) A and R both are correct statements but R is not the correct explanation of the A
 c) A is true, but R is false
 d) A is false , but R is true



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Answers For A Ls Chemistry

**Dr. Benvikram Barman, Dr. Bhupendra
Singh Banjare, Dr. Dolly Baghel, Dr.
Dinesh Kumar Tandan, Mr. Naveen
Verma, Mr. Rupendra Kumar Sahu**

Answers For A Ls Chemistry:

Reviewing **Answers For A Ls Chemistry**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Answers For A Ls Chemistry**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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