

Calculations in Analytical Chemistry

The Mole

The mole is the SI unit for amount of substance. It corresponds to the amount of a chemical substance that contains the same number of particles present in 12 grams of C-12 which is equivalent to Avogadro's number, that is, $N_A = 6.022 \times 10^{23}$.

The molar mass is given as the mass of a compound in grams per mole of that substance. It is calculated by adding the molar masses of all the component elements in the compound. Take glucose for example:

Element	Molar mass per element	Number of atoms	Mass
C	12 g/mol	6	72 g/mol
H	1 g/mol	12	12 g/mol
O	16 g/mol	6	96 g/mol
Molar Mass			180 g/mol

Mole and Millimole Calculations

The millimole is simply 1/1000 of a mole. It is sometimes used in calculations because of its convenience.

Example (Skoog, et al., 2014):

1. Find the number of moles and millimoles of benzoic acid (122.1 g/mol) that are contained in 2.00 g of pure acid.

If we use HBz to represent benzoic acid:

$$HBz_{mol} = 2.00 \text{ g}_{HBz} \times \frac{1 \text{ mol}_{HBz}}{122.1 \text{ g}_{HBz}} = 0.0164 \text{ mol}_{HBz}$$

To obtain the number of millimoles, the millimolar mass of benzoic acid will be used:

$$HBz_{mol} = 2.00 \text{ g}_{HBz} \times \frac{1 \text{ mmol}_{HBz}}{0.1221 \text{ g}_{HBz}} = 16.4 \text{ mmol}_{HBz}$$

2. What is the mass in grams of Na^+ (22.99 g/mol) in 25.0 g of Na_2SO_4 (142.0 g/mol)?

$$\begin{aligned} \text{mass}_{Na^+} &= 25.0 \text{ g}_{Na_2SO_4} \times \frac{1 \text{ mol}_{Na_2SO_4}}{142.0 \text{ g}_{Na_2SO_4}} \times \frac{2 \text{ mol}_{Na^+}}{1 \text{ mol}_{Na_2SO_4}} \times \frac{22.99 \text{ g}_{Na^+}}{1 \text{ mol}_{Na^+}} \\ \text{mass}_{Na^+} &= 8.10 \text{ g}_{Na^+} \end{aligned}$$

Analytical Chemistry Calculation Questions And Answers

John Kenkel



Analytical Chemistry Calculation Questions And Answers:

Analytical Chemistry for Technicians John Kenkel, 2002-10-29 Surpassing its bestselling predecessors this thoroughly updated third edition is designed to be a powerful training tool for entry level chemistry technicians *Analytical Chemistry for Technicians Third Edition* explains analytical chemistry and instrumental analysis principles and how to apply them in the real world A unique feature of this edition is that it brings the workplace of the chemical technician into the classroom With over 50 workplace scene sidebars it offers stories and photographs of technicians and chemists working with the equipment or performing the techniques discussed in the text It includes a supplemental CD that enhances training activities The author incorporates knowledge gained from a number of American Chemical Society and PITTCON short courses and from personal visits to several laboratories at major chemical plants where he determined firsthand what is important in the modern analytical laboratory The book includes more than sixty experiments specifically relevant to the laboratory technician along with a Questions and Problems section in each chapter *Analytical Chemistry for Technicians Third Edition* continues to offer the nuts and bolts of analytical chemistry while focusing on the practical aspects of training **Basic Analytical Chemistry** Mr. Rohit Manglik, 2024-05-18 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels

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analysis instrumental analysis spectrochemical methods such as atomic spectroscopy and UV Vis and IR molecular spectrometry chromatographic techniques including gas chromatography and high performance liquid chromatography electroanalytical methods and more Incorporating an additional ten years of teaching experience since the publication of the third edition the author has made significant updates and enhancements to the fourth edition More than 150 new photographs and either new or reworked drawings spanning every chapter to assist the visual learner A new chapter on mass spectrometry covering GC MS LC MS LC MS MS and ICP MS Thirteen new laboratory experiments An introductory section before chapter 1 to give students a preview of general laboratory considerations safety laboratory notebooks and instrumental analysis Additional end of chapter problems expanded report type questions and inclusion of relevant section headings in the Questions and Problems sections Application Notes in each chapter An appendix providing a glossary of quality assurance and good laboratory practice GLP terms

Basic Analytical Chemistry (Penerbit USM) Faiz Bukhari Mohd Suah, Wan Saime Wan Ngah, Che Sofiah Saidin, 2018 BASIC ANALYTICAL CHEMISTRY Malaysia is a fast developing country Realizing the need to provide experts in chemistry this book is appropriate to be used as a text for fundamental course in analytical chemistry The texts cover topics from the most basic analytical chemistry course including methods on basic analyses to important concepts such as handling of data analysis chemical equilibrium stoichiometry and titration The chemical equilibrium in this book covers acid base equilibrium precipitation complex and redox titration For every topic examples and solutions are provided to give reader a better understanding in the topics covered

Analytical Chemistry Gary D. Christian, Purnendu K. Dasgupta, Kevin A. Schug, 2013-10-07 With the 7th Edition of Analytical Chemistry renowned chemists Purnendu Sandy Dasgupta and Kevin Schug both of the University of Texas Arlington join the author team The new edition focuses on more in depth coverage of the principles and techniques of quantitative analysis and instrumental analysis aka Analytical Chemistry The goal of the text is to provide a foundation of the analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry

Fundamentals of Biochemical Calculations Krish Moorthy, 2007-11-30 Fundamentals of Biochemical Calculations Second Edition demystifies the fundamental calculations used in modern biochemistry cell biology and allied biomedical sciences The book encourages both undergraduates and scientists to develop an understanding of the processes involved in performing biochemical calculations rather than rely on memory

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completely new approach to learning and teaching the fundamentals of analytical chemistry It summarizes 250 basic concepts of the field on the basis of slides Each of the nine chapters offers the following features Introduction Summary General scheme Teaching objectives Text containing the explanation of each slide Recommended and commented bibliography Questions to be answered Slides A distinct feature of this novel book is its focus on the fundamental concepts and essential principles of analytical chemistry which sets it apart from other books presenting descriptive overviews of methods and techniques **Report** Great Britain. Dept. of Science and Art, 1869

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