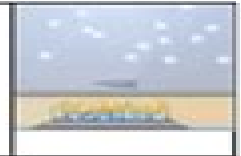


Activity C:

Altitude and phase changes

Get the Gizmo ready:

- Click **Reset**.
- Set **Ice volume** to 0 cm³.
- Set the **Altitude** to 5,000 meters (16,404 feet).



Question: The **altitude** of a location is its vertical distance above sea level. How does altitude affect phase changes?

1. **Form a hypothesis:** As altitude increases, the air pressure decreases. How do you think the lower pressure will affect the following? (Highlight your answers. ✎)

A. Freezing point:	Increase	Stay the same	Decrease
B. Melting point:	Increase	Stay the same	Decrease
C. Boiling point:	Increase	Stay the same	Decrease

2. **Experiment:** Use the Gizmo to find the freezing, melting, and boiling points of water at 5,000 meters (16,404 feet). Write these values below.

Freezing point: Melting point: Boiling point:

3. **Analyze:** How did altitude affect the freezing, melting, and boiling points of water?

Boiling point dropped, while freezing and melting points remained unchanged.

4. **Challenge:** Try to explain these results based on the fact that air pressure decreases with altitude. If possible, discuss your ideas with your classmates and teacher.

Because there is less pressure on it, breaking the bonds requires less kinetic energy when there is less air pressure.

5. **Apply:** Why does pasta take longer to cook in the mountains?

The pasta cooks more slowly because the water can't reach a high enough temperature.

6. **Apply:** A pressure cooker allows food to be cooked under high pressure. Why is this useful?

It cooks more quickly under increased pressure because it reaches a higher temperature.

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Table of Contents 1 3 Climate Changes Answer Key

1. Understanding the eBook 1 3 Climate Changes Answer Key
 - The Rise of Digital Reading 1 3 Climate Changes Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying 1 3 Climate Changes Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 1 3 Climate Changes Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from 1 3 Climate Changes Answer Key
 - Personalized Recommendations
 - 1 3 Climate Changes Answer Key User Reviews and Ratings
 - 1 3 Climate Changes Answer Key and Bestseller Lists
5. Accessing 1 3 Climate Changes Answer Key Free and Paid eBooks

- 1 3 Climate Changes Answer Key Public Domain eBooks
- 1 3 Climate Changes Answer Key eBook Subscription Services
- 1 3 Climate Changes Answer Key Budget-Friendly Options
- 6. Navigating 1 3 Climate Changes Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - 1 3 Climate Changes Answer Key Compatibility with Devices
 - 1 3 Climate Changes Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 1 3 Climate Changes Answer Key
 - Highlighting and Note-Taking 1 3 Climate Changes Answer Key
 - Interactive Elements 1 3 Climate Changes Answer Key
- 8. Staying Engaged with 1 3 Climate Changes Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 1 3 Climate Changes Answer Key
- 9. Balancing eBooks and Physical Books 1 3 Climate Changes Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 1 3 Climate Changes Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 1 3 Climate Changes Answer Key
 - Setting Reading Goals 1 3 Climate Changes Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 1 3 Climate Changes Answer Key
 - Fact-Checking eBook Content of 1 3 Climate Changes Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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