Buffers Keep the Balance

ata Table 1. Model Carbonate Blood Buffer

| Table 1 | Test Tube Number | | | | | | |
|--|---------------------------------|------------------------------------|--------------------|------------------------------------|--------------------|--------------------------------------|--|
| | 1 | 2 | 3 | 4 | 5 | 6 | |
| Solution | Carbonic Acid (Reference) | Model Carbonate Blood Buffer | Water (Control) | Model Carbonate Blood Buffer | Water (Control) | Sodium Bicarbonate (Reference) | |
| Bromthymol blue indicator color | Yenow . | green 8h | green 600 | men 13h | 9 Heen 600 | bive | |
| Initial pH value | 6.0 | 7.2 | 6.0 | 7.0 | 6.0 | 8.0 | |
| Number of drops of HCl required to convert solutions 2 and 3 to yellow (acid reference color) | NA | 16 | 5 | NA | NA | NA | |
| Number of drops of NaOH required to convert solutions 4 and 5 to blue "hase reference color) | NA | NA | NA | 2 | 2 | NA | |

Data Table 2. Effect of HCl on Biological Phosphate Buffers

| | Estimated pH (pH paper) | Universal Indicator Color | | | | |
|-----------------------|----------------------------|---------------------------|------------------------|-------------------------|--------------------------|--|
| | | Initial | After 1 drop of HCl | After 5 drops of HCI | After 10 drops of HCl | |
| Water (control) | 6.0 | 6.0 | 6.0 | NA | NA | |
| Phosphate Buffer A | 6.0 | 60 | 6.0 | 6.0 | | |
| Phosphate Buffer B | 6.4 | 6.4 | 6.4 | 6.0 | 6.0 | |

Data Table 3. Effect of NaOH on Biological Phosphate Buffers

| | | Universal Indicator Color | | | | |
|-----------------------|------------|---------------------------|-------------------------|--------------------------|--------------------------|--|
| | (pH paper) | Initial | After 1 drop of NaOH | After 5 drops of NaOH | After 10 drops of HCI | |
| (control) | 6.0 | 6.0 | 7.0 | NA | NA NA | |
| Phosphate Buffer A | 6.0 | 6.0 | 6.8 | 9 . | 200 | |
| Phosphate Buffer B | 6.4 | 6.4 | 8.0 | 9.0 | 8.0 | |

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Beth Reekles

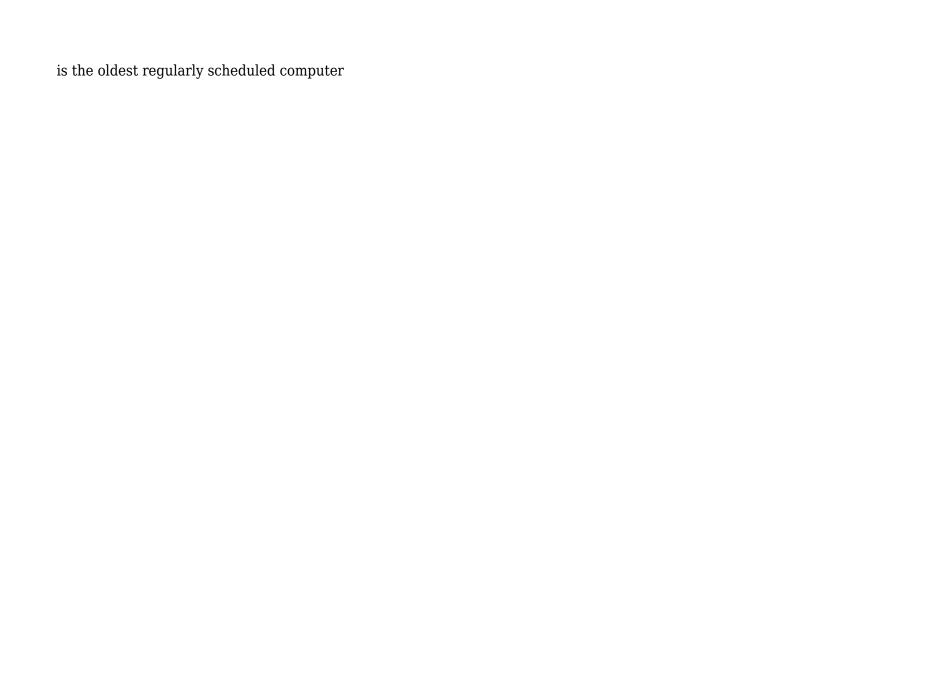
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