

**Types of Chemical Reaction Worksheet**

Balance the reactions **1 to 6** and indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented:

1.  $2 \text{NaBr} + \text{Ca(OH)}_2 \rightarrow \text{CaBr}_2 + 2 \text{NaOH}$  Reaction Type : single displacement
2.  $2 \text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow 3 (\text{NH}_4)_2\text{SO}_4$  Reaction Type : synthesis
3.  $4 \text{C}_3\text{H}_8\text{O} + 27 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$  Reaction Type : combustion
4.  $3 \text{Pb} + 2 \text{H}_3\text{PO}_4 \rightarrow 3 \text{H}_2 + \text{Pb}_3(\text{PO}_4)_2$  Reaction Type : single displacement
5.  $\text{Li}_3\text{N} + 3 \text{NH}_4\text{NO}_3 \rightarrow 3 \text{LiNO}_3 + (\text{NH}_4)_3\text{N}$  Reaction Type : double displacement
6.  $\text{HBr} + 3 \text{Al(OH)}_3 \rightarrow \text{H}_2\text{O} + 3 \text{AlBr}_3$  Reaction Type : double displacement

Indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented in 7 to 20.

7.  $\text{Na}_3\text{PO}_4 + 3 \text{KOH} \rightarrow 3 \text{NaOH} + \text{K}_3\text{PO}_4$  Reaction Type **double displacement**
8.  $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2 \text{LiCl}$  Reaction Type **double displacement**
9.  $\text{C}_8\text{H}_{12} + 9 \text{O}_2 \rightarrow 6 \text{CO}_2 + 6 \text{H}_2\text{O}$  Reaction Type **combustion**
10.  $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$  Reaction Type **single displacement**
11.  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$  Reaction Type **decomposition**
12.  $\text{P}_4 + 3 \text{O}_2 \rightarrow 2 \text{P}_2\text{O}_3$  Reaction Type **synthesis**
13.  $2 \text{RbNO}_3 + \text{BeF}_2 \rightarrow \text{Be(NO}_3)_2 + 2 \text{RbF}$  Reaction Type **double displacement**
14.  $2 \text{AgNO}_3 + \text{Cu} \rightarrow \text{Cu(NO}_3)_2 + 2 \text{Ag}$  Reaction Type **single displacement**
15.  $\text{C}_3\text{H}_8\text{O} + 4 \text{O}_2 \rightarrow 3 \text{CO}_2 + 3 \text{H}_2\text{O}$  Reaction Type **combustion**
16.  $2 \text{C}_2\text{H}_5 + \text{Fe} \rightarrow \text{Fe(C}_2\text{H}_5)_2$  Reaction Type **synthesis**
17.  $\text{SeCl}_6 + \text{O}_2 \rightarrow \text{SeO}_2 + 3 \text{Cl}_2$  Reaction Type **single displacement**
18.  $2 \text{MgI}_2 + \text{Mn(SO}_4)_2 \rightarrow 2 \text{MgSO}_4 + \text{MnI}_4$  Reaction Type **double displacement**
19.  $\text{O}_3 \rightarrow \text{O} + \text{O}_2$  Reaction Type **decomposition**
20.  $2 \text{NO}_2 \rightarrow 2 \text{O}_2 + \text{N}_2$  Reaction Type **decomposition**

# Answers For Chemical Reactions

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**Chemical Reactions, Materials and Particles** Gill Murphy, 2006-03 Specials are a teaching resource designed for KS3 students whose literacy skills are considerably lower than their age. These books have an older format to counteract the simple text and cover various topics. They include activities, visuals, and assessment sheets as well as teacher pages.

**Regents Exams and Answers: Chemistry--Physical Setting Revised Edition** Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers: Chemistry provides essential practice for students taking the Chemistry Regents, including actual recently administered exams and thorough answer explanations for all questions. This book features eight actual administered Regents Chemistry exams so students can get familiar with the test. Thorough explanations for all answers, self-analysis charts to help identify strengths and weaknesses, test-taking techniques and strategies, a detailed outline of all major topics tested on this exam, a glossary of important terms to know for test day.

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**The Handy Engineering Answer Book** DeLean Tolbert Smith, Aishwary Pawar, Nicole P. Pitterson, Debra-Ann C. Butler, 2022-09-20 A handy resource on the fundamental facts about engineering for both engineers and non engineers alike whether you are exploring engineering for the first time already have a strong background or fall anywhere in between Engineering impacts every aspect of our lives Bridges buildings buses electrical grids computers televisions refrigerators vacuum cleaners and virtually any everyday household item needs to be engineered to function properly Fundamentally engineering is about identifying a need and

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