
Chem - Covalent Bonding: Questions & Answers

properties of non metals ✓ Ans - inability to conduct electricity - absence of ions or delocalised electrons
low boiling and melting points - generally weak intermolecular bonds, can be soft

molecules ✓ Ans - separate groups of atoms with a formula, bonded together

intermolecular bonds ✓ Ans - bonds between molecules (dispersion, dipole-dipole, hydrogen bonding) determines physical properties

intramolecular bonds ✓ Ans - bonds between atoms in a molecule (covalent, ionic eg.)

covalent bonds ✓ Ans - non metals share electrons to complete the octet rule, an atom forms as many bonds as it needs eg. carbon needs 4 electrons so forms 4 bonds

why do non metals share electrons not transfer? ✓ Ans - non metals have a relatively high electronegativity so competing to attract each others electron, shares instead

electron dot diagrams ✓ Ans - lewis diagram shows valence electrons (bonding pairs) and lone pairs

valence structure ✓ Ans - each pair of bonding electrons is shown as a line, still represents dots as lone pairs. 1 line=2 electrons

bonding pairs and lone pairs ✓ Ans - bonding pairs; shared electrons in bond
lone pairs; non bonding pairs

double covalent bond ✓ Ans - rigid, two pairs of electrons (4 electrons) are shared, represented as a double line in valence structures eg. O_2

Answers For Chem Think Covalent Bonding

Leallyn Burr Clapp



Answers For Chem Think Covalent Bonding:

Reviewing **Answers For Chem Think Covalent Bonding**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Answers For Chem Think Covalent Bonding**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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