

Unit 3 Review - Atomic Structure & Periodic Table

Part 1: Answer the following questions.

1. a. Which scientist created the first modern atomic theory? John Dalton
 b. What was his theory?
 1. All matter is made of extremely small particles called atoms.
 2. All atoms of a given element are identical and atoms of a specific element are different from any other element
 3. Atoms cannot be created, divided into smaller particles or destroyed
 4. Different atoms combine in simple whole number ratios to form compounds
 5. In a chemical reaction, atoms are separated, combined or rearranged.
- c. What was his model of the atom called? Billiard Ball
- d. What was proven wrong and why?
 1. All atoms of a given element are identical – wrong b/c of the existence of isotopes
 2. Atoms cannot be divided into smaller particles – wrong b/c of the existence of electrons, protons and neutrons
2. a. Which scientist did the gold foil experiment? Rutherford
 b. What did he conclude from the gold foil experiment?
 1. Atoms are made up of mostly empty space
 2. Atoms have a small positive center call the nucleus
3. a. Who discovered the electron? J. J. Thomson
 b. What was the name of his experiment? Cathode Ray Tube Experiment
 c. What was his model of the atom called? Plum Pudding Model
4. What did Mendeleev do?
Arranged the periodic table in order of increasing atomic mass
5. a. Who arranged the **modern** periodic table? Moseley
 b. How is the modern periodic table arranged? Arranged in order of Atomic Number
6. a. What are the three subatomic particles? Give the symbols, charge, mass and location of each.

Particles	Symbol	Charge	Mass	Location
Proton	p^+	Positive	1 amu	Nucleus
Neutron	n^0	Neutral	1 amu	Nucleus
Electron	e^-	Negative	1/2000 amu	Electron Cloud

7. What is special about the proton? The proton IDENTIFIES THE ELEMENT
8. Fill in the blanks in the following:
 - a. Mass # = $p^+ + n^0$
 - b. Atomic # = # of protons
9. What is the difference between Average Atomic mass and mass #? (Average) Atomic Mass is found on the periodic table and is the weighted average of the isotopes. The Mass Number is the mass of one element and must be given or calculated by adding the protons and neutron of that element.
10. Which of the following are isotopes of the same element? How do you know? A, and D. They have the same Atomic Number (# of Protons)
 - a. $^{12}_5X$
 - b. $^{15}_5X$
 - c. $^{12}_6X$
 - d. $^{15}_7X$
11. Calculate the average atomic mass for Boron if it has two naturally occurring isotopes: boron -10 (abundance of 19.8%) and boron - 11 (abundance of 80.2%).

$$\left(10 \times \frac{19.8}{100} \right) + \left(11 \times \frac{80.2}{100} \right) = 10.802 \text{ amu}$$
12. Identify an element "X" by calculating the average atomic mass. X = 6 (abundance of 7.5%) and X= 7 (abundance 92.5%).

$$\left(6 \times \frac{7.5}{100} \right) + \left(7 \times \frac{92.5}{100} \right) = 6.925 \text{ amu} \quad \text{Which is Lithium}$$

Answer Key Unit 3 Review Electrons

William J. Callaway



Answer Key Unit 3 Review Electrons:

Mosby's Radiation Therapy Study Guide and Exam Review Leia Levy, 2025-08-21 Reinforce your understanding of radiation therapy and prepare for the Registry exam Mosby's Radiation Therapy Study Guide and Exam Review Second Edition is both a study companion for Washington and Leaver's Principles and Practice of Radiation Therapy and a superior review for the ARRT Radiation Therapy Certification Exam This completely updated edition reflects the latest exam specifications and features an easy to read format that presents information in concise bullets and tables More than 2 000 total multiple choice questions in Registry format provide a realistic testing experience to prepare you for the real exam NEW Quality control procedures and guidelines for linear accelerators NEW Fractionation and protraction considerations UPDATED Content reflects the latest ARRT Radiation Therapy Certification Exam Specifications UPDATED Radiation Protection and Safety and Overview of Cancer Imaging and Management Modalities chapters offer the most current information in these key areas EXPANDED Charge capture and record keeping content enhances the Oncology Patient Care chapter EXPANDED Additional cross sectional images provide anatomy review and reinforce treatment planning concepts More than 2 000 multiple choice questions in Registry format in the text and on the companion Evolve website provide a realistic exam experience Complete coverage helps you prepare for the ARRT Radiation Therapy Certification Exam Content review in outline and tabular format provides a concise recap of the material you need to know to succeed on the exam Exercises at the end of each section offer engaging active review opportunities *Jacaranda Chemistry 2 VCE Units 3 and 4, 3e learnON and Print* Robert Stokes, Angela Stubbs, Neale Taylor, Jason Bourke, Ben Williams, Maida Derbogolian, 2023-12-26 Developed by expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027 **Mosby's Comprehensive Review of Radiography - E-Book** William J. Callaway, 2012-02-22 A complete review for the Registry exam Mosby's Comprehensive Review of Radiography The Complete Study Guide and Career Planner 6th Edition covers the five major subject areas of the ARRT exam in radiography It is also an effective study guide for many radiography courses Written in outline format each review of a subject is followed by questions related specifically to that area Two mock ARRT exams are included in the book and online exams include a pool of over 1 400 review questions that may be randomly combined to generate a virtually limitless number of mock ARRT exams From noted radiography educator William J Callaway this edition also provides advice on writing resumes and cover letters interviewing employer expectations and continuing education requirements to help you make the transition to a successful career Review of the five major subject areas covered on the ARRT exam in an outline format helps you concentrate on the most important information Over 2 400 review questions in the book and online offer practice with a multiple choice format similar to the ARRT exam Thorough coverage of digital and computed radiography

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Chapter Resource 3 Cell Structure Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004

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Holt Chemistry Ralph Thomas Myers, 2004 **CUET (UG) 20 Combined Mock Test Papers Science (PCM) Stream (English, Physics, Chemistry, Mathematics, General Test) For 2025 Exam** Oswaal Editorial Board, 2024-09-05

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web oct 8 2020 this list is based on score only so it doesn't reflect the difference between a non typical and a typical but to make it easy we listed from lowest to highest score these are the top 12 mule deer bucks that were featured in the 2021 monster mule deer calendar see the current monster mule deer calendar here

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