

ANSWER KEY: BUILD AN ATOM

PART I: ATOM SCREEN

Build an Atom simulation (<http://phet.colorado.edu/en/simulation/build-an-atom>)

1. Explore the *Build an Atom* simulation with your group. As you explore, talk about what you find.
2.
 - a) List two things your group observed in the simulation.
Responses will vary, but here is what you should see....
 - Adding protons changes the identity of the atom - adding one moves from right to left on the Periodic Table.
 - You can get the atom to be stable/unstable by adjusting the number of neutrons
 - Adding electrons will change it from a +ion to neutral atom, to -ion depending on how many electrons you add
 - Adding electrons - you can only add 2 to the first orbital and 8 to the second
 - b) What particle(s) are found in the center of the atom?
Protons and neutrons
3. Play until you discover which **particle(s)** determine(s) the name of the **element** you build. What did you discover?
Protons - adding them moves from left to right on the Periodic Table
4. What is the **name** of the following atoms?
 - a) An atom with 3 protons and 4 neutrons: **Lithium**
 - b) An atom with 2 protons and 4 neutrons: **Helium**
 - c) An atom with 4 protons and 4 neutrons: **Beryllium**
5. Play with the simulation to discover which particles affect the **charge** of an atom or ion.
 - a) Fill in the blanks below to show your results:

Neutral atoms have the same number of protons and electrons.

Positive ions have more protons than electrons.

Negative ions have fewer protons than electrons.
 - b) Develop a relationship (in the form of a single sentence or equation) that can predict the charge based on the number and types of particle.
The charge of an atom is determined by the ratio of protons to electrons.

Answers For Build An Atom Phet

Graeme Lofts, Merrin J. Evergreen



Answers For Build An Atom Phet:

This book delves into Answers For Build An Atom Phet. Answers For Build An Atom Phet is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Answers For Build An Atom Phet, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Answers For Build An Atom Phet
 - Chapter 2: Essential Elements of Answers For Build An Atom Phet
 - Chapter 3: Answers For Build An Atom Phet in Everyday Life
 - Chapter 4: Answers For Build An Atom Phet in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Answers For Build An Atom Phet. The first chapter will explore what Answers For Build An Atom Phet is, why Answers For Build An Atom Phet is vital, and how to effectively learn about Answers For Build An Atom Phet.
 3. In chapter 2, this book will delve into the foundational concepts of Answers For Build An Atom Phet. The second chapter will elucidate the essential principles that need to be understood to grasp Answers For Build An Atom Phet in its entirety.
 4. In chapter 3, this book will examine the practical applications of Answers For Build An Atom Phet in daily life. The third chapter will showcase real-world examples of how Answers For Build An Atom Phet can be effectively utilized in everyday scenarios.
 5. In chapter 4, this book will scrutinize the relevance of Answers For Build An Atom Phet in specific contexts. The fourth chapter will explore how Answers For Build An Atom Phet is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Answers For Build An Atom Phet. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Answers For Build An Atom Phet.

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Answers For Build An Atom Phet Introduction

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