

# Chemistry Study Guide

### Periodic Table of the Elements

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| <div>1<br/>H</div>   |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                       |                       |                       |                       | <div>18<br/>Ar</div>  | <div>19<br/>K</div>   | <div>20<br/>Ca</div>  |                       |                       |                       |                       |                       |                       |                       |                       |                       |                       | <div>36<br/>Kr</div>  | <div>37<br/>Rb</div>  | <div>38<br/>Sr</div> |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      | <div>54<br/>Xe</div> | <div>55<br/>Cs</div> | <div>56<br/>Ba</div> |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                     | <div>86<br/>Rn</div> |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                     |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |                      |
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## 🔍 Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🧪

## 💡 Matter:

Anything that has **mass** and **occupies space** 🧪🌐

Examples: air, gases, liquids, solids!

## 🔬 Atoms:

Atoms are the **smallest, indivisible particle** 🧪👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## 📌 Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location    |
|----------|--------|------------|--------|-------------|
| Proton   | $p^+$  | 1          | +1     | Nucleus     |
| Neutron  | $n$    | 1          | 0      | Nucleus     |
| Electron | $e^-$  | $\sim 0$   | -1     | Orbit/Shell |

# Answers For Holt Chemistry Study Guide

**AW Chickering**



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