

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Balancing Chemical Equations

**Vocabulary:** coefficient, combination, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

The scouts are making s'mores out of toasted marshmallows, chocolate, and graham crackers.

1. What is wrong with the image below? \_\_\_\_\_



2. Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, how many s'mores could you make with the ingredients shown? \_\_\_\_\_

### Gizmo Warm-up

In a chemical reaction, **reactants** interact to form **products**. This process is summarized by a chemical equation. In the *Balancing Chemical Equations* Gizmo™, look at the floating molecules below the initial reaction:  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ .



- How many atoms are in a hydrogen **molecule** ( $\text{H}_2$ )? \_\_\_\_\_
- How many atoms are in an oxygen molecule ( $\text{O}_2$ )? \_\_\_\_\_
- How many hydrogen and oxygen atoms are in a water molecule ( $\text{H}_2\text{O}$ )? \_\_\_\_\_
- In general, what does a **subscript** (such as the "2" in  $\text{H}_2$ ) tell you about the molecule?  
\_\_\_\_\_
- A chemical equation is **balanced** if the number of each type of atom on the left side is equal to the number of each type on the right side. Is this reaction balanced? \_\_\_\_\_

# Answers For Student Exploration Chemical Equations Answers

**Patrick Brown**



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