#### Native the opasitive into expressions (for 16) for the following reactions:

#### For each of the following, write the equilibrium expression and salve for K. Show all world?

 PCL(g) → PCL(g) + CL(g). What is the equilibrium constant if the equilibrium consentrations are as follows: PCL is 6:0896 mol L, PCh is 0.0247 mol L and Ch is 0.0247 mol L?

7. As a given interpretation, the K<sub>in</sub> for the resistant (Hittg) → H<sub>2</sub>(g) + I<sub>2</sub>(g) is 1.40 × 10<sup>-1</sup>. If the consentration of tests H<sub>1</sub> and I<sub>1</sub> as equilibrium are 2.00 × 10<sup>-1</sup>M. find the consequentials of MI.

#### In each problem, calculate the soluting concentration or estented at equilibrium.

	13600	1396	10.0	Man
86.7	0.78	6-173	10(3.702)	No.
100	34	0.242	0.242	0.247
100	C-90/T905	0.112	200	1. 12 May 12 May 71

# **1chemistry Worksheet Answers**

**ML Morrison** 

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