

Chemistry 105 Practice Exam 1

Multiple Choice Section

1. An element _____

- a. can be separated into its components by physical methods.
- b. always has the same chemical properties regardless of its source.
- c. cannot be separated into simpler substances by chemical methods.
- d. can also be a compound.
- e. exists only as atoms, not as molecules.

2. Which one of the following represents a chemical change?

- a. Mercury (II) oxide is heated to and forms mercury metal and oxygen gas.
- b. Rubbing alcohol evaporates.
- c. Iodine vapor deposits on a surface.
- d. Iron metal is separated from sand using a magnet.
- e. Rock salt is pulverized.

3. Equal amounts of water are present under the following conditions. In which case do the water molecules have the highest kinetic energy?

- a. as ice at -10°C (14°F)
- b. as steam at 100°C (212°F)
- c. in the liquid phase at 80°C (176°F)
- d. in the gas phase at 150°C (302°F)
- e. in the solid phase at 0°C (32°F)

4. Which of the following represents the largest mass?

- a. 250 ng
- b. 25 μg
- c. 2.5 g
- d. 0.25 kg
- e. 25 mg

5. A student conducts repeated trials to determine the density of a sample of seawater and obtains the following results: 1.321 g/mL, 1.323 g/mL, 1.319 g/mL, and 1.321 g/mL. Known values are reported to be in the 1.02 to 1.09 g/mL range. The experimental results are _____

- a. less precise than the known values.
- b. more accurate than the known values.
- c. imprecise but accurate.
- d. precise and accurate.
- e. inaccurate but precise.

6. According to the law of definite proportions, _____

- a. atoms forming a given compound react in variable proportions depending on conditions.
- b. different samples of the same compound contain the same proportions of the same elements.
- c. all compounds containing the same types of atoms have identical properties.
- d. all compounds containing the same types of atoms have relative masses that are whole-number multiples.
- e. only one type of molecule can be produced when two elements combine.

7. Which of the following statements is NOT true regarding blackbody radiation, the ultraviolet catastrophe, and Planck's view of electromagnetic radiation?

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**Neale Taylor, Angela Stubbs, Robert
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Table of Contents 1practice Problems Chemistry

1. Understanding the eBook 1practice Problems Chemistry
 - The Rise of Digital Reading 1practice Problems Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying 1practice Problems Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 1practice Problems Chemistry
 - User-Friendly Interface

4. Exploring eBook Recommendations from 1practice Problems Chemistry
 - Personalized Recommendations
 - 1practice Problems Chemistry User Reviews and Ratings
 - 1practice Problems Chemistry and Bestseller Lists
5. Accessing 1practice Problems Chemistry Free and Paid eBooks
 - 1practice Problems Chemistry Public Domain eBooks
 - 1practice Problems Chemistry eBook Subscription Services
 - 1practice Problems Chemistry Budget-Friendly Options
6. Navigating 1practice Problems Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - 1practice Problems Chemistry Compatibility with Devices
 - 1practice Problems Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 1practice Problems Chemistry
 - Highlighting and Note-Taking 1practice Problems Chemistry
 - Interactive Elements 1practice Problems Chemistry
8. Staying Engaged with 1practice Problems Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 1practice Problems Chemistry
9. Balancing eBooks and Physical Books 1practice Problems Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 1practice Problems Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 1practice Problems Chemistry
 - Setting Reading Goals 1practice Problems Chemistry
 - Carving Out Dedicated Reading Time

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 - Fact-Checking eBook Content of 1practice Problems Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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