

## 3 - 4 Review and Reinforcement

### Elements and Compounds

Fill in each blank with the name of the element.

| Elemental Matter | Elemental Compound | Elementary Matter |
|------------------|--------------------|-------------------|
| 1. Sodium        | NaCl               | Sulfur            |
| 2. Magnesium     |                    |                   |
| 3. Lead          | Pb                 | Carbon            |
| 4. Oxygen        | O <sub>2</sub>     | Hydrogen          |
| 5. Iron          | Fe                 | Helium            |
| 6. Zinc          |                    |                   |
| 7. Potassium     | K                  |                   |
| 8. Copper        | Cu                 |                   |
| 9. Aluminum      | Al                 |                   |
| 10. Nitrogen     | N <sub>2</sub>     |                   |
| 11. Carbon       | C                  |                   |

Complete the following by inserting the missing words appropriate to each blank line from the list.

element  
compound  
mixture

elementary  
compound  
mixture

elementary matter  
matter  
mixture (blend)

1. Elements are matter composed of identical atoms.
2. Elements are elementary matter, which cannot be broken down.
3. Elements are composed of pure particles.
4. The smallest elementary particles are called atoms or molecules.
5. Mixture is a blend of two or more elements or compounds.
6. Elements and compounds are called pure substances because they have a composition and physical properties.

## 2 3 Review And Reinforcement Matter Answers

**S. Rao Vallabhaneni**



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