

# CHM 111 Analysis of Aluminum-Zinc Alloy using Ideal Gas Law

## Report Sheet

### Data Sheet:

	Trial 1	Trial 2
Mass of gelatin capsule	<u>1.7382</u> g	<del>_____</del> g
Mass of alloy sample plus capsule	<u>1.9852</u> g	<del>_____</del> g
Mass of empty beaker	<u>104.3024</u> g	<del>_____</del> g
Mass of beaker plus displaced water	<u>346.225</u> g	<del>_____</del> g
Barometric pressure	_____ mm Hg	_____ mm Hg
Temperature	<u>24</u> °C	_____ °C
Calculations Mass of alloy sample	<u>0.247</u> g	<del>_____</del> g
Mass of displaced water	<u>241.9226</u> g	<del>_____</del> g
Volume of displaced water	_____ mL	<del>_____</del> mL
Volume of H <sub>2</sub> , V	_____ L	<del>_____</del> L
Temperature of H <sub>2</sub> , T	<u>297.15</u> K	_____ K
Vapor pressure of water at T	<u>22.4</u> mm Hg	_____ mm Hg
Pressure of dry H <sub>2</sub> ,	<u>755.7</u> mm Hg; <u>0.9943</u> atm	_____ mm Hg; _____ atm
Moles H <sub>2</sub> from sample,	_____ moles	<del>_____</del> moles
Moles H <sub>2</sub> per gram of sample,	_____ moles/g	<del>_____</del> moles/g %
Al (read from graph)	_____ %	<del>_____</del> %
Unknown no. <u>3</u>		

# Analysis Of Aluminum Zinc Alloy Lab Answers

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## **Table of Contents Analysis Of Aluminum Zinc Alloy Lab Answers**

1. Understanding the eBook Analysis Of Aluminum Zinc Alloy Lab Answers
  - The Rise of Digital Reading Analysis Of Aluminum Zinc Alloy Lab Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Analysis Of Aluminum Zinc Alloy Lab Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Analysis Of Aluminum Zinc Alloy Lab Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Analysis Of Aluminum Zinc Alloy Lab Answers
  - Personalized Recommendations
  - Analysis Of Aluminum Zinc Alloy Lab Answers User Reviews and Ratings

- Analysis Of Aluminum Zinc Alloy Lab Answers and Bestseller Lists
- 5. Accessing Analysis Of Aluminum Zinc Alloy Lab Answers Free and Paid eBooks
  - Analysis Of Aluminum Zinc Alloy Lab Answers Public Domain eBooks
  - Analysis Of Aluminum Zinc Alloy Lab Answers eBook Subscription Services
  - Analysis Of Aluminum Zinc Alloy Lab Answers Budget-Friendly Options
- 6. Navigating Analysis Of Aluminum Zinc Alloy Lab Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Analysis Of Aluminum Zinc Alloy Lab Answers Compatibility with Devices
  - Analysis Of Aluminum Zinc Alloy Lab Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Analysis Of Aluminum Zinc Alloy Lab Answers
  - Highlighting and Note-Taking Analysis Of Aluminum Zinc Alloy Lab Answers
  - Interactive Elements Analysis Of Aluminum Zinc Alloy Lab Answers
- 8. Staying Engaged with Analysis Of Aluminum Zinc Alloy Lab Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Analysis Of Aluminum Zinc Alloy Lab Answers
- 9. Balancing eBooks and Physical Books Analysis Of Aluminum Zinc Alloy Lab Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Analysis Of Aluminum Zinc Alloy Lab Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Analysis Of Aluminum Zinc Alloy Lab Answers
  - Setting Reading Goals Analysis Of Aluminum Zinc Alloy Lab Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Analysis Of Aluminum Zinc Alloy Lab Answers
  - Fact-Checking eBook Content of Analysis Of Aluminum Zinc Alloy Lab Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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