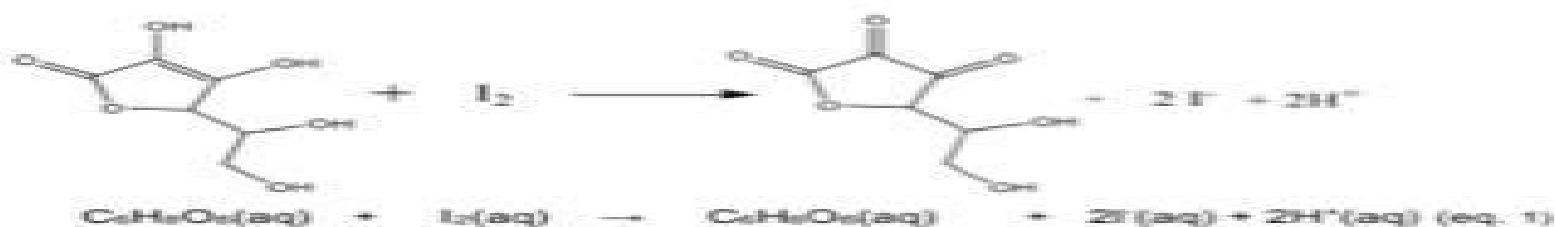


Volumetric Analysis for Vitamin C

The labelling and quantification of active ingredients in drugs is strictly controlled in Canada under the Food and Drugs Act. You can be fairly confident when you buy a prescription drug or over-the-counter medication that the listed ingredients are present in their indicated quantities. Vitamins and minerals are regulated as a sub-set of drugs and are governed by the Natural Health Products Regulation. While the regulations assure quality control, and require that companies report accurate and detailed information about the safety, quality, and efficacy of their products, the government does not have the resources to check quality control in the plethora of supplements on the market. During manufacture of large batches of pills, for example, problems may arise with ensuring a homogenous distribution of active ingredients. A recent study from the US (where regulations are more lax) found a wide variance in how much vitamin D was present in supplements compared to what was reported on the bottle.¹ A study such as this indicates the importance of analytical methods in determining how much of the active ingredient is truly present as compared to the label.

This experiment will quantitatively analyze an over-the-counter Vitamin C tablet for the mass of Vitamin C. Iodine reacts with Vitamin C by an oxidation process as indicated by the following reaction equation:



Note that one mole of iodine reacts with one mole of Vitamin C. In this experiment, a solution of Vitamin C (made from the tablet) is added to a solution containing excess iodine (with known concentration). Vitamin C is the limiting reactant; all of the Vitamin C reacts and leaves some iodine unreacted. The amount of iodine in excess is determined by titration with the standardized sodium thiosulfate from Lab D. Since the total (initial) amount of iodine was known, and the amount of excess iodine was determined in the titration, the amount of iodine that reacted with vitamin C can be determined by the difference. Thus, the amount of vitamin C can be determined.

The process of titrating an excess amount of a reactant not consumed in a reaction is called a 'back titration'. The related equations from Lab D are reproduced here for reference:



1. Leblanc, E.S.; Perrin, N.; Johnson, J.D. Over-the-Counter and Compounded Vitamin D: Is Potency What We Expect? *JAMA Intern Med.* **2013**, 173, 585-586.

Analysis Of Vitamin C Lab Answers

Minjie Lin



Analysis Of Vitamin C Lab Answers:

Unveiling the Power of Verbal Artistry: An Emotional Sojourn through **Analysis Of Vitamin C Lab Answers**

In a global inundated with monitors and the cacophony of immediate communication, the profound energy and mental resonance of verbal artistry often disappear into obscurity, eclipsed by the continuous assault of noise and distractions. Yet, nestled within the musical pages of **Analysis Of Vitamin C Lab Answers**, a captivating perform of literary beauty that pulses with organic emotions, lies an unique journey waiting to be embarked upon. Published by way of a virtuoso wordsmith, that magical opus manuals readers on an emotional odyssey, lightly revealing the latent possible and profound impact stuck within the complicated internet of language. Within the heart-wrenching expanse of the evocative analysis, we shall embark upon an introspective exploration of the book is central styles, dissect their captivating publishing type, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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