

24. Which of the following is an aldehyde?



All none of the above

EO 1

EO 2

EO 3

EO 4

25. Trends in the physical properties of 2-propanone can be explained by which of the following intermolecular forces?

- A) London Dispersion Forces
 B) Hydrogen bonding
 C) dipole-dipole interactions
 D) ionic bonding
 E) none of the above

26. Which of the following has the highest boiling point?

- A) CH_4 B) $\text{CH}_3\text{CH}_2\text{OH}$ C) CH_3COCH_3 D) $\text{CH}_3\text{CH}_2\text{CH}_3$
 EO 1 EO 2 EO 3

27. Which of the following compounds is expected to be the most soluble in water?

- A) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$
 B) CH_4
 C) $\text{CH}_3\text{CH}_2\text{CH}_2\text{OCH}_2\text{CH}_3$
 D) $\text{CH}_3\text{CH}_2\text{OH}$
 E) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$

28. Pick out the oxygen-containing molecule with the lowest boiling point.



EO 1

EO 2

EO 3

EO 4 and 5

29. The molecular formula for heptane is

- A) C_7H_{16} B) C_7H_{14} C) C_7H_{12} D) C_7H_{10} E) C_7H_8

30. The primary product of the following reaction is which of the following?



31. Given the balanced equation representing a reaction. This organic reaction is best classified as



- A) is addition reaction
 B) is elimination reaction
 C) is polymerization reaction
 D) is substitution reaction

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