

Name:

Class: VIII

Science Preparatory Material

1. A student sets up an experiment to test the conductivity of different liquids. She finds that distilled water does not conduct electricity, but saltwater does.

Explain why this happens and suggest how the student can turn distilled water into a conductor.

2. During an experiment, a student notices bubbles forming around the electrodes when an electric current passes through a salt solution. Describe the process happening in the solution and explain why bubbles are forming near the electrodes.
3. Which of the following liquids is the best conductor of electricity?

- (A) Distilled water
- (B) Tap water
- (C) Vegetable oil
- (D) Sugar solution

4. Consider the following passage:

"When an electric current passes through a conducting solution, it causes chemical reactions. These reactions can result in the deposition of metals or the formation of gas bubbles at the electrodes."

Based on this passage, explain the role of electrodes in electrolysis and provide an example of a metal that can be deposited through this process.

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