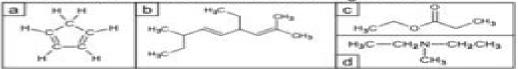
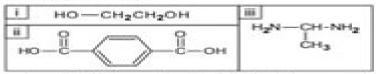
Organic chemistry review

- 1. Draw structures for the following compounds:
 - a) 2-methylbutane
 - b) 3,3-dichlorohexane
 - c) 4-ethyl-3,3,4-trimethyldecane
 - d) ethanamide
 - e) cis-1,3-dimethylcyclohexane

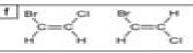
- f) 2-methyl-4-octanol
- g) propanal
- h) methyl ethanoate
- i) methyl propyl ether
- j) 2-pentanone
- Write the correct name for each of the following structures:



- Draw the polymers that result from:
 - a) the reaction of i) and ii)
 - b) the reaction of ii) and iii)
 - c) the addition of ethene molecules
 - d) the addition of 3-hexene molecules



- 4. What is the difference between a saturated and unsaturated hydrocarbon? Which is benzene?
- Identify each pair as structural isomers, geometric isomers, or neither
 - a) cyclopentane, pentane
 - b) 1,1-dichloroethene, trans-1,2-dichloroethene
 - c) hexanoic acid, propyl propanoate
 - d) cis-1,2-dichlorocyclopentane, trans-1,2-dichlorocyclopentane
 - e) cis-1,2-dichlorocyclopentane, trans-1,3-dichlorocyclopentane



- C₂H₆O exists as two structural isomers. Draw each isomer and name each. Predict the relative boiling point and solubility in water of the two isomers. Explain.
- 7. Indicate if the functional group contains a cabonyl, carboxyl or hydroxyl group:

	alcohol	ketone	ester	carboxylic acid
carboxyl group				
carbonyl group				
hydroxyl group				

- In each case indicate how the following molecule can be made by drawing the reaction, naming all products and reactants, and naming the type of reaction that is occurring.
 - a) 1,2-dichlorocyclopentane (from a hydrocarbon)
 - b) 1,1-dichlorocyclopentane (from a hydrocarbon)
 - A condensation reaction to make butyl methanoate
 - d) The reactants from c) (not using hydrogenation)
- 9. Using Br₂(aq), how can you easily distinguish between pentane, 1-pentene, and 1-pentyne?
- 10. What is wrong with the following names?
 - a) 3-methyl-2-butyne
 - b) 1,2-dichlorocyclobutane
 - c) 3-methyloctene

- d) 2,3-dimethyl-4-ethylnonane
- e) 1-butanone
- f) 2-ethyl-2-methylhexane

Additional practice (pg. 1049-1053): 24.5, 7, 11, 12, 13, 14, 17, 19, 32, 36, 37

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Michael B. Smith, Jerry March

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