

Chemistry 12 - Review of Significant Digits

TOTAL 28

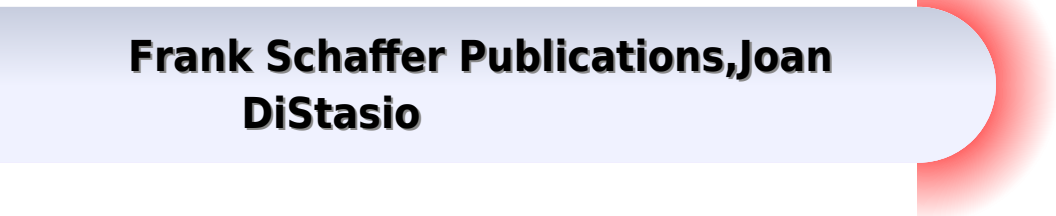
The rules for zeros in significant digits are as follows:

- All zeros between non-zero digits are significant.
 - Zeros at the beginning of a number (eg. 0.0095) are NOT SIGNIFICANT!
If the number 0.0095 was written in scientific notation, it would be: 9.5×10^{-3} . The exponent is not counted as significant so this number has 2 significant digits.
 - Zeros on the right side of a number (at the end) are significant if the DECIMAL POINT is shown.
eg) 50.00 has 4 significant digits
43.0 has 3 significant digits
20. has 2 significant digits
100. has 3 significant digits
 - Zeros to the left of an UNDERSTOOD decimal point are NOT significant.
eg) 300 has 1 significant digit
10 000 has 1 significant digit
12 320 has 4 significant digits
420 has 2 significant digits
1. Find the number of significant digits in each of the following measurements:

- | | |
|----------------------------------|-----------------------------------|
| a) 3.4005 5 | f) 9.080×10^{-3} 4 |
| b) 2580 3 | g) 1.00 3 |
| c) 3.20×10^{-2} 3 | h) 0.0027890 5 |
| d) 0.000308 3 | i) 320 000 2 |
| e) 23.000 5 | j) 9 1 |

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