

The Periodic Table of the Elements

1 H 1.01												18 He 4.00					
3 Li 6.94	4 Be 9.01											13 B 10.81	14 C 12.01	15 N 14.01	16 O 16.00	17 F 18.99	18 Ne 20.18
11 Na 22.99	12 Mg 24.31											19 K 39.10	20 Ca 40.08				
37 Rb 85.47	38 Sr 87.62											55 Cs 132.91	56 Ba 137.33	57-70 Lanthanides			
87 Fr (223)	88 Ra (226)	89-102 Actinides										113 Uut (284)	114 Uuq (289)	115 Uup (288)	116 Uuh (293)	117 Uus (294)	118 Uuo (294)

Alkali metals

Alkaline earth metals

Transition metals

Other metals

Metalloids (semi-metals)

Nonmetals

Halogens

Noble gases

Four Earth Elements

Element name

Mercury

Atomic #

80

Symbol

Hg

Avg. Mass

200.59

*lanthanides

**actinides

57 La 138.91	58 Ce 140.12	59 Pr 140.91	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.97	64 Gd 157.25	65 Tb 158.93	66 Dy 162.50	67 Ho 164.93	68 Er 167.26	69 Tm 168.93	70 Yb 173.04
89 Ac (227)	90 Th 232.04	91 Pa 231.04	92 U 238.03	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)

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**National Institute for Occupational
Safety and Health**



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The Periodic Table I D. Michael P. Mingos, 2020-02-05 As 2019 has been declared the International Year of the Periodic Table it is appropriate that Structure and Bonding marks this anniversary with two special volumes In 1869 Dmitri Ivanovitch Mendeleev first proposed his periodic table of the elements He is given the major credit for proposing the conceptual framework used by chemists to systematically inter relate the chemical properties of the elements However the concept of periodicity evolved in distinct stages and was the culmination of work by other chemists over several decades For example Newland s Law of Octaves marked an important step in the evolution of the periodic system since it represented the first clear statement that the properties of the elements repeated after intervals of 8 Mendeleev s predictions demonstrated in an impressive manner how the periodic table could be used to predict the occurrence and properties of new elements Not all of his many predictions proved to be valid but the discovery of scandium gallium and germanium represented sufficient vindication of its utility and they cemented its enduring influence Mendeleev s periodic table was based on the atomic weights of the elements and it was another 50 years before Moseley established that it was the atomic number of the elements that was the fundamental parameter and this led to the prediction of further elements Some have suggested that the periodic table is one of the most fruitful ideas in modern science and that it is comparable to Darwin s theory of evolution by natural selection proposed at approximately the same time There is no doubt that the periodic table occupies a central position in chemistry In its modern form it is reproduced in most undergraduate inorganic textbooks and is present in almost every chemistry lecture room and classroom This first volume provides chemists with an account of the historical development of the Periodic Table and an overview of how the Periodic Table has evolved over the last 150 years It also illustrates how it has guided the research programmes of some distinguished chemists

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3 Building Blocks Catalysis and Coupling Chemistry Volume 4 Protection Reactions Medicinal Chemistry Combinatorial Synthesis Volume 5 Analysis and Function of Amino Acids and Peptides The fourth volume in this series is structured in three main sections The first section is about protection reactions and amino acid based peptidomimetics The second and most extensive part is devoted to the medicinal chemistry of amino acids It includes among others the chemistry of alpha and beta amino acids peptide drugs and advances in N and O glycopeptide synthesis The final part deals with amino acids in combinatorial synthesis Methods such as phage display library peptide synthesis and computational design are described Originally planned as a six volume series Amino Acids Peptides and Proteins in Organic Chemistry now completes with five volumes but remains comprehensive in both scope and coverage Further information about the 5 Volume Set and purchasing details can be viewed here [United States Congressional Serial Set, Serial No. 14795, House Reports Nos. 741-771](#) Congress, House,

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