



Name: _____
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KEY

Nuclear Radiation Worksheet

1. a. What determines whether an atom is stable or unstable?

Its overall size and a difference between proton # and neutron #

- b. How does an unstable atom gain stability?

It undergoes radioactive decay

- c. What part of an atom undergoes change during radioactive decay?

The nucleus!

2. Which of the human senses can detect radiation?

Our eyes can detect colors, but we have no senses to detect alpha, beta, or gamma radiation

3. Explain the difference between an isotope and a radioactive isotope.

An isotope has a different # of protons & neutrons. Radioactive isotope is larger, unstable, and has a different # of p

4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	4 also 4 neutrons	Positive (+)	Stopped by paper or paper
Beta	0 Neutrons (electron)	Negative (-)	Stopped by sheets of tin foil or thick magazine
Gamma	0 Neutrons	Neutral	Stopped by several inches of lead or feet of concrete

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	-4	-2
Beta	No change	+1
Gamma	No change	No change

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.



7. Complete the following nuclear equations:



Answer Key For Chemistry Radioactive Decay Worksheet

Xiang Xie



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Table of Contents Answer Key For Chemistry Radioactive Decay Worksheet

1. Understanding the eBook Answer Key For Chemistry Radioactive Decay Worksheet
 - The Rise of Digital Reading Answer Key For Chemistry Radioactive Decay Worksheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Chemistry Radioactive Decay Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Chemistry Radioactive Decay Worksheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Chemistry Radioactive Decay Worksheet
 - Personalized Recommendations
 - Answer Key For Chemistry Radioactive Decay Worksheet User Reviews and Ratings
 - Answer Key For Chemistry Radioactive Decay Worksheet and Bestseller Lists
5. Accessing Answer Key For Chemistry Radioactive Decay Worksheet Free and Paid eBooks

- Answer Key For Chemistry Radioactive Decay Worksheet Public Domain eBooks
- Answer Key For Chemistry Radioactive Decay Worksheet eBook Subscription Services
- Answer Key For Chemistry Radioactive Decay Worksheet Budget-Friendly Options
- 6. Navigating Answer Key For Chemistry Radioactive Decay Worksheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key For Chemistry Radioactive Decay Worksheet Compatibility with Devices
 - Answer Key For Chemistry Radioactive Decay Worksheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key For Chemistry Radioactive Decay Worksheet
 - Highlighting and Note-Taking Answer Key For Chemistry Radioactive Decay Worksheet
 - Interactive Elements Answer Key For Chemistry Radioactive Decay Worksheet
- 8. Staying Engaged with Answer Key For Chemistry Radioactive Decay Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key For Chemistry Radioactive Decay Worksheet
- 9. Balancing eBooks and Physical Books Answer Key For Chemistry Radioactive Decay Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key For Chemistry Radioactive Decay Worksheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key For Chemistry Radioactive Decay Worksheet
 - Setting Reading Goals Answer Key For Chemistry Radioactive Decay Worksheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key For Chemistry Radioactive Decay Worksheet
 - Fact-Checking eBook Content of Answer Key For Chemistry Radioactive Decay Worksheet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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