

u/may/23

two hydrogen atoms share their valence electrons, both atoms achieve the electronic configuration of noble gas & satisfy the duplet rule.



Types of Covalent Bond:

Depending upon the number of bond pair, a covalent bond is further classified into three types.

- ◊ Single Covalent Bond
- ◊ Double Covalent Bond
- ◊ Triple Covalent Bond

Single Covalent Bond (-)

A covalent bond which is formed by the mutual sharing of one bond pair is called a single covalent bond and it is represented by a single short straight line. The formation of $\text{H}-\text{H}$, $\text{H}-\text{Cl}$, CH_4 are few examples of this type of bonding.

Examples:

Formation of Chlorine Molecule:

One more electron to achieve a stable outer electronic configuration. When two chlorine atoms share their valence

Answer Key For Convelent Bond Caoncept Map

Minjie Lin



Answer Key For Convelent Bond Caoncept Map:

A-level Chemistry E. N. Ramsden,2000 Each topic is treated from the beginning without assuming prior knowledge Each chapter starts with an opening section covering an application These help students to understand the relevance of the topic they are motivational and they make the text more accessible to the majority of students Concept Maps have been added which together with Summaries throughout aid understanding of main ideas and connections between topics Margin points highlight key points making the text more accessible for learning and revision Checkpoints in each chapter test students understanding and support their private study **Holt Chemistry** Ralph Thomas Myers,2004 **Organic Chemistry** T. W. Graham Solomons,Craig B. Fryhle,Scott A. Snyder,2023 Organic Chemistry 13th edition provides a comprehensive yet accessible treatment of all the essential organic chemistry concepts with emphasis on relationship between structure and reactivity in the subject The textbook includes all the concepts covered in a typical organic chemistry textbook but is unique in its skill development approach to the subject Numerous hands on activities and real world examples are integrated throughout the text to help students understand both the why and the how behind organic chemistry This International Adaptation offers new and updated content with improved presentation of all course material It offers new material on several topics including the relevance of intermolecular forces in the immune response and vaccines like those for Covid 19 the chemistry of breathing carbonic anhydrase how conjugation and complexation affect the color of lobsters and how biodegradable polymers are used to stabilize vaccines and pharmaceuticals Content is revised to reflect the current understanding of chemical processes and improved depictions of longstanding mechanisms This edition builds on the ongoing pedagogical strength of the book with the inclusion of additional worked and end of chapter problems and an engaging set of new problems entitled Chemical Consultant Needed These draw from the primary chemical literature and give students experience of working with more complex polyfunctional structures and areas where key transformations take place

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