

Prelab

Enzymatic Decomposition and Analysis of Hydrogen Peroxide

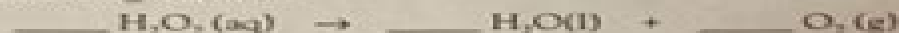
Name _____

Date _____

Section _____

1. What gas is being collected in today's experiment?

2. Balance the following reaction:



3. In today's experiment you determine that you collected 1.33×10^{-3} mols of oxygen. Calculate (show your setup for the conversion!) how many mols of H_2O_2 were in your hydrogen peroxide solution. Note: You will need a conversion factor from the equation you balanced in question 2.

_____ mols of H_2O_2

4. Based on your answer above, how many grams of H_2O_2 were in the hydrogen peroxide solution? (H_2O_2 has a molecular weight of 34.0 g/mol)

_____ grams of H_2O_2

5. Assume the grams of H_2O_2 calculated in question 4 were found in a 6.00 gram sample of a hydrogen peroxide solution. Calculate the mass percent H_2O_2 in the solution.

_____ % H_2O_2

Analysis Of Hydrogen Peroxide Prelab Answers

Terry C. Jones



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