

5. Carbonyl chloride, COCl_2 , is a colourless gas. It is made by the following reaction.



- (a) When the pressure on the equilibrium mixture is decreased, the position of equilibrium moves to left.
- (i) How does the concentration of each of the three chemicals change?
- _____
- _____ [2]
- (ii) Explain why the position of equilibrium moves to left.
- _____
- _____ [2]
- (b) Using the information given with the equation, is the forward reaction exothermic or endothermic? Give a reason for your choice.
- _____
- _____ [2]
- (c) Carbonyl chloride reacts with water to form two acidic compounds. Suggest which acidic compounds are formed.
1. _____
2. _____ [2]
- (d) The structural formula of carbonyl chloride is given below.



Draw a diagram that shows the arrangement of the valency electrons in one molecule of this covalent compound.

Use x for an electron from a chlorine atom.

Use o for an electron from a carbon atom.

Use • for an electron from an oxygen atom.

[4]
[Total: 12]

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Table of Contents 0620 31 M J 08 Ms

1. Understanding the eBook 0620 31 M J 08 Ms
 - The Rise of Digital Reading 0620 31 M J 08 Ms
 - Advantages of eBooks Over Traditional Books
2. Identifying 0620 31 M J 08 Ms
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 0620 31 M J 08 Ms
 - User-Friendly Interface
4. Exploring eBook Recommendations from 0620 31 M J 08 Ms
 - Personalized Recommendations
 - 0620 31 M J 08 Ms User Reviews and Ratings
 - 0620 31 M J 08 Ms and Bestseller Lists
5. Accessing 0620 31 M J 08 Ms Free and Paid eBooks
 - 0620 31 M J 08 Ms Public Domain eBooks
 - 0620 31 M J 08 Ms eBook Subscription Services
 - 0620 31 M J 08 Ms Budget-Friendly Options
6. Navigating 0620 31 M J 08 Ms eBook Formats

- ePub, PDF, MOBI, and More
 - 0620 31 M J 08 Ms Compatibility with Devices
 - 0620 31 M J 08 Ms Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 0620 31 M J 08 Ms
 - Highlighting and Note-Taking 0620 31 M J 08 Ms
 - Interactive Elements 0620 31 M J 08 Ms
 8. Staying Engaged with 0620 31 M J 08 Ms
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 0620 31 M J 08 Ms
 9. Balancing eBooks and Physical Books 0620 31 M J 08 Ms
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 0620 31 M J 08 Ms
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine 0620 31 M J 08 Ms
 - Setting Reading Goals 0620 31 M J 08 Ms
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of 0620 31 M J 08 Ms
 - Fact-Checking eBook Content of 0620 31 M J 08 Ms
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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