

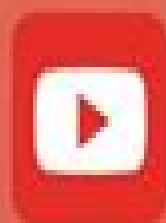
IGCSE A* MODEL ANSWERS

0620 CHEMISTRY

MAY/JUNE - 2024

PAPER - 61

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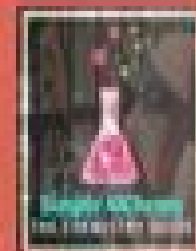
Table 1 shows the tests and the student's observations for solid P.

Tests	Observations
Test 1 Heat about half of solid P in boiling tube with Bunsen burner flame.	<u>disappears + effs</u> The white solid forms a colourless liquid, steam effervesces out from the boiling tube and condensation is seen at the top of the boiling tube. After a while the colourless liquid becomes a white solid.
Test 2 The remaining solid P is dissolved in water to form solution P. Solution P is divided into three portions. To the first portion of solution P, a few drops of sodium hydroxide solution are added and then it is warmed. Repeat the procedure and then using red litmus paper at the mouth of the boiling tube. Test 3 To the second portion of solution P, a few drops of dilute hydrochloric acid are added and then it is warmed. Test 4 To the third portion of solution P, with a few drops of distilled aqueous potassium manganate(VII).	<u>no precipitate</u> <u>no effs</u> <u>no effs</u> <u>no change</u> The liquid red litmus paper remains red. <u>white precipitate</u> <u>SO₄²⁻</u> The solution becomes light purple. <u>SO₄²⁻</u>

3. (a) The observations in test 1 show that solid P is hydrated.

Describe a chemical test to show that the condensation at the top of the boiling tube contains water.

anhydrous blue cobalt chloride paper
turns pink
anhydrous white crumbly solid



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S Nieto



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