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IGCSE A* MODEL ANSWER

0620 CHEMISTRY

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Paper 62



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- (b) Add a suitable scale to the y-axis in Fig. 2.1. Your scale should extend to 2°C above your highest temperature in Table 2.1.

Plot your results from Experiments 1 to 6 on the grid.

Draw two straight lines through your points, one through the first three points, and one through the last three points. Extend your straight lines so that they cross.

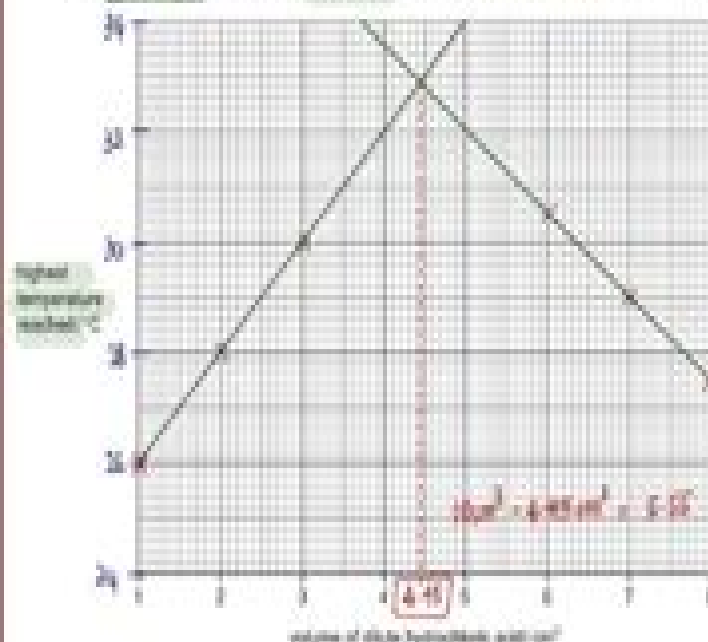


Fig. 2.1

11

- (c) The point on the graph where the two straight lines cross is where all of the aqueous sodium hydroxide reacts with all of the dilute hydrochloric acid to form a neutral solution.

- (i) Use your graph in Fig. 2.1 to deduce the volume of dilute hydrochloric acid and the volume of aqueous sodium hydroxide that react together to produce a neutral solution. Show your working on Fig. 2.1.

volume of dilute hydrochloric acid = 4.4 cm^3 ↓

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VM Jensen



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Table of Contents 0620 62 O N 14markscheme

1. Understanding the eBook 0620 62 O N 14markscheme
 - The Rise of Digital Reading 0620 62 O N 14markscheme
 - Advantages of eBooks Over Traditional Books
2. Identifying 0620 62 O N 14markscheme
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 0620 62 O N 14markscheme
 - User-Friendly Interface
4. Exploring eBook Recommendations from 0620 62 O N 14markscheme
 - Personalized Recommendations
 - 0620 62 O N 14markscheme User Reviews and Ratings

- 0620 62 0 N 14markscheme and Bestseller Lists
- 5. Accessing 0620 62 0 N 14markscheme Free and Paid eBooks
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 - 0620 62 0 N 14markscheme eBook Subscription Services
 - 0620 62 0 N 14markscheme Budget-Friendly Options
- 6. Navigating 0620 62 0 N 14markscheme eBook Formats
 - ePub, PDF, MOBI, and More
 - 0620 62 0 N 14markscheme Compatibility with Devices
 - 0620 62 0 N 14markscheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 0620 62 0 N 14markscheme
 - Highlighting and Note-Taking 0620 62 0 N 14markscheme
 - Interactive Elements 0620 62 0 N 14markscheme
- 8. Staying Engaged with 0620 62 0 N 14markscheme
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 0620 62 0 N 14markscheme
- 9. Balancing eBooks and Physical Books 0620 62 0 N 14markscheme
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 0620 62 0 N 14markscheme
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 0620 62 0 N 14markscheme
 - Setting Reading Goals 0620 62 0 N 14markscheme
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 0620 62 0 N 14markscheme
 - Fact-Checking eBook Content of 0620 62 0 N 14markscheme
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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