

- 4 A student investigated the reaction between aqueous potassium manganate(VII), which is purple, and two different colourless acidic solutions, **D** and **E**.

For Examiner's Use

Three experiments were carried out.

(a) Experiment 1

A burette was filled with the solution of potassium manganate(VII) to the 0.0 cm³ mark. Using a measuring cylinder, 25 cm³ of solution **D** was poured into a conical flask.

Potassium manganate(VII) solution was added to the flask until the mixture just turned permanently pink.

Use the burette diagram to record the final volume in the table and complete the table.



final reading

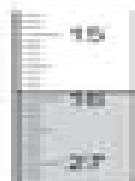
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initial burette reading / cm ³	
difference / cm ³	

[2]

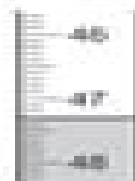
(b) Experiment 2

Experiment 1 was repeated using 25 cm³ of solution **E** instead of solution **D**.

Use the burette diagrams to record the readings in the table and complete the table.



initial reading



final reading

	burette reading
final burette reading / cm ³	
initial burette reading / cm ³	
difference / cm ³	

[2]

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Richard Bailey

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