

Inorganic and Physical Chemistry Unit***Electromagnetic radiation and atomic spectra******Section A Questions*****Electromagnetic Radiation**

1 A tunable laser emits blue light. If the wavelength is increased

- A the velocity increases
- B the frequency increases
- C the wavenumber increases
- D the colour changes towards red.

SQP4

2 Which of the following lists electromagnetic radiation bands in order of increasing frequency?

- A Ultraviolet, visible, infra-red, radio
- B Radio, infra-red, visible, ultraviolet
- C Radio, microwave, ultraviolet, visible
- D Visible, ultraviolet, X-ray, microwave

2013Rev2

3 The energy associated with a photon of electromagnetic radiation is

- A independent of the frequency
- B proportional to the frequency
- C inversely proportional to the frequency
- D proportional to the square of the frequency.

2014Rev6

Atomic Spectra

4 Sodium vapour street lamps emit yellow light because

- A sodium vapour is burning and giving out a yellow glow
- B sodium vapour filters out all the light from the filament except yellow
- C energy corresponding to yellow light is given out as electrons in sodium move to lower energies
- D energy corresponding to yellow light is given out as electrons in sodium move to higher energies.

SQP5

5 Using information from the Data Booklet which one of the following metal salts will emit radiation of the highest frequency when placed in a Bunsen flame?

- A Copper(II) sulfate
- B Potassium chloride
- C Barium chloride
- D Lithium sulfate

2013Rev3

6 Which of the following indicators transmits only the lower frequencies of the visible spectrum at low pH?

Indicator	Colour in acid	Colour in alkali
A	Violet	Red
B	Green	Blue
C	Yellow	Violet
D	Red	Yellow

2013Rev4

1 Advanced Higher Chemistry Past Paper

JG Myers



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