11-2 Practice Problems

 Determine the mass of lithium hydroxide produced when 0.38 g of lithium nitride reacts with water according to the following equation:

Eglin x 1 mai Liju x 3 mai Liot x 24 g Liot = 0.78 g 1 x 35 g Liju x 1 mai Liot x 1 mai Liot Liot Liot

> What mass of sodium chloride is produced when chlorine reacts with 0.29 g of sodium iodide?

029 NoI + Cl2 - 2NoCI + Iz 029 NoI + Cl2 - 2NoCI + Iz 029 NoI x 1mol x 2mol NoI (mol NoC) NoCI

> Determine the mass of carbon dioxide produced when 0.85 g of butane reacts with oxygen according to the following equation:

0+85gCyHo x 1mol x 5mol Cox x 44g = 2.58g 1 x 58g x 2mol CyHo 1mol CO z

 Determine the mass of antimony produced when 0.46 g of antimony(III) oxide reacts with carbon according to the following equation:

0.4695bD= x [mol x 2mol 5b, 03 x 121.769 fa.389 Sb

5. What mass of hydrogen peroxide (H₂O₂)
must decompose to produce 0.77 g of
water?
2H₂O₂ -> 2H₂O + O₂

0.779 H20 x (mol x 2420 x 349 = [.4594:02

6. What mass of carbon monoxide must react with oxygen to produce 0.69 g of carbon dioxide?

0.69gCOz × 1mol × 2mol CO x 289 = 0.449 1 449 2mol COz 1mol = 0.449 Determine the mass of sodium nitrate produced when 0.73 g of nickel(II) nitrate reacts with sodium hydroxide according to the following equation:

 $\frac{N8(NO_3)_2 + 2N8OH \rightarrow N8(OH)_2 + 2N8NO_3}{O.73_3 V80_3)_2 \times \frac{1000}{185_3} \times \frac{2001 N8NO_3}{1001 N6NO_3)_2} \times \frac{85_3}{1001} = \frac{(2.68_3)}{1001 N6NO_3}$

 Determine the mass of calcium hydroxide produced when calcium carbide reacts with 0.64 g of water according to the following equation:

 $0.64_9H_2O \times \frac{CaC_2 + 2H_2O - Ca(OH)_2 + C_2H_2}{189} \times \frac{1001(a(0H)_2 + C_2H_2)}{1001} \times \frac{749}{1001} = \frac{1.329}{Ca(0H)_2}$

How many grams of ozone (O₃) must decompose to produce 0.87 g of oxygen?

0.8790 x 1mg/ x 2mg/ x 48903 = 0.87909

 Find the mass of sugar (C₆H₁₂O₆) required to produce 1.82 L of carbon dioxide gas at STP from the reaction described by the following equation: C₆H₁₂O₆ → 2C₂H₆O + 2CO₂

 How many liters of oxygen are necessary for the combustion of 425 g of sulfur, assuming that the reaction occurs at STP? The balanced equation is S + O₂ → SO₂.

 Find the mass of benzene (C₆H₆) required to produce 2.66 L of carbon dioxide gas at STP from the reaction described by the following equation:

2C4H6 + 15O2 → 6H2O + 12CO2

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Stanley E. Manahan

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The Impact of Climate Change on Our Life Abdelnaser Omran, Odile Schwarz-Herion, 2018-02-23 This book introduces the highly topical issue from many different angles sensitizing readers to the various challenges to human life posed by climate change identifying possible intentional and inadvertent anthropogenic factors and consequences and seeking socially and environmentally viable solutions. The book begins by examining the impact of the climate change discussion on science politics economy and culture from its historical origin in the first Club of Rome Report and its inclusion in the UN s SDGs to the Paris Agreement and beyond Comprising 12 chapters it analyses the factors which caused the catastrophic 2014 Kelantan flood in Malaysia focusing on the Kuala Krai district and discusses mud architecture in Wadi Hadramout Yemen and mitigating the expected effects of climate change on this unique architecture and cultural heritage It also examines the economic costs of climate change on health and the increased burden on individual expenditures and national health systems. The role of climate change in the water energy nexus and efforts to increase efficiency in energy and water end use to increase Queensland's agricultural sector's resilience in Australia is addressed as is water security and climate change issues in developing countries and the potential of partnership procurement strategies for managing sustainable urban water supply in Nigerian cities It also includes a chapter offering a new approach to waste management exploring to what extent waste can complicate our daily actions and influence environmental decay and recommending that renewable materials be sorted and separated from other types of materials to avoid cross contamination to increase the value of the materials and to ease the process of manufacturing Subsequent chapters identify factors sustaining the municipal solid waste management and practices in Ajdabiya city in Libya and look at accounting disclosure remedies by exploring areas in which sustainability reporting could expand beyond corporate environmental reporting to additional disclosures curbing recklessness in pursuing merely economic goals The book shows from the perspective of agriculture how human activities can increase the negative impacts of climate change on lifestyle in Malaysia suggesting alternative lifestyles and encouraging international cooperative efforts The last chapters evaluate the impacts of various environmental factors on the local tourism sector in Pakistan and discuss strategies to tackle climate change focusing on the opportunities and risks of climate engineering Since these risks encompass inadvertent negative effects and targeted abuse for covert weather warfare and terrorism that violate the UN s ENMOD convention the author recommends viable alternatives to deal with climate change

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