- 2. Assertion: Chips and other snacks become rancid due to oxidation.
 - Chips and other snacks become to the snacks they are packed in cartes Reason:

Answers 1.3

- (b) It is due to formation of Ag_aS.
 - (d) He or N₂ because these are inert gases.
 - 3. (b) CuCO₃.Ču(OH)₂ is green.
 - (d) Fe₂O₃xH₂O is the chemical formula of reddish brown rust.
 - 5. (c) BHT (Butylated hydroxy toluene)
- 1. (a) Both 'A' and 'R' are true and 'R' is correct explanation of the assertion.
 - (c) 'A' is true but 'R' is false. Foods are packed in inert gases like nitrogen to prevent from rancidity.

CASE STUDY QUESTIONS

 A student took HCl in a conical flask and was placed on a white tile with cross mark. When seen through the flask the mark was visible. On adding some sodium thiosulphate in the flask the cross mark disappeared when seen through the flask. The student observed some reaction as the products formed were different than the reactants. Both the reactants were transparent.



- (i) Name the type of reaction seen in the set up.
- (a) Combustion (b) Combination (c) Precipitation (ii) Why did the cross mark disappear?
 - (a) Decomposition
 - (a) The solution formed in the flask does not allow the light to pass through it to see the cross mark
 - (b) The sulphur is a colloid in this reaction, staying in suspension and eventually blocking the light from reaching the solution
 - (c) The solution transforms from being colourless to being milky and entirely opaque (d) All of the above
- (iii) Select the correct reaction for the above chemical reaction
 - $(a) \ \operatorname{Na_2S_2O_3}\left(\operatorname{aq}\right) + 2\operatorname{HCl}\left(\operatorname{aq}\right) \longrightarrow 2\operatorname{NaCl}\left(\operatorname{aq}\right) + \operatorname{H_2O(l)} + \operatorname{SO_2}\left(\operatorname{g}\right) + \operatorname{S(s)}$
 - (b) $S_2O_3^{-2}$ (aq) + 2H* (aq) \longrightarrow H₂O (l) + SO₂ (g) + S(s)
 - (c) $Na_2S_2O_3$ (aq) + 2HCl (aq) \longrightarrow 2NaCl (aq) + SO_2 (g) + S(8)
 - (d) None of the above
- (iv) Name the resultant insoluble substance formed in the flask
 - (a) Copper(II) iodide (c) Sulphur
- (v) What type of reaction occurs in the flask?
 - (a) Displacement reaction
- (c) Photo-chemical decomposition Ans. (i) (c)(ii) (cl)(iii) (a)
- (b) Potassium hydroxide
- (d) Sodium Thiosulphate
- (b) Double displacement reaction
- (d) Thermal decomposition
- (iv) (c)(v) (b)

2014 Msce Questions Physical Science

R. G. Reid

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International Conference on Mechanism Science and Control Engineering (MSCE 2014), 2014-09-02 The aim of MSCE 2014 is to provide a platform for researchers engineers and academicians as well as industrial professionals to present their research results and development activities in mechanism science and control engineering It provides opportunities for the delegates to exchange new ideas and application experiences to establish business or research relations and to find global partners for future collaboration MSCE2014 is conducted to all the researchers engineers industrial professionals and academicians who are broadly welcomed to present their latest research results academic developments or theory practice Topics of interest include but are not limited to Mechanism theory and Application Mechanical control and Automation Engineering Mechanical Dynamics Materials Processing and Control Instruments and Vibration Control It is of great pleasure to see the delegates exchanging ideas and establishing sound relationships on the conference Peripheral Nerve Injury An Anatomical and Physiological Approach for Physical Therapy Intervention Stephen Carp, 2015-04-21 Here's everything you need to know about peripheral nerve injuries and how to recognize and treat acute and chronic injuries and conditions across the lifespan In depth discussions organized in a streamlined format ensure you understand the identification pathophysiology assessment and procedural interventions associated with peripheral nerve injuries Build the knowledge base you need to evaluate the most common to complex injuries make a diagnosis and implement a plan of care with this one of a kind Environmental Engineering for the 21st Century National Academies of Sciences, Engineering, and resource Medicine, National Academy of Engineering, Division on Engineering and Physical Sciences, Division on Earth and Life Studies, Water Science and Technology Board, Ocean Studies Board, NAE Office of Programs, Board on Life Sciences, Board on Environmental Studies and Toxicology, Board on Earth Sciences and Resources, Board on Energy and Environmental Systems, Board on Chemical Sciences and Technology, Board on Atmospheric Sciences and Climate, Board on Agriculture and Natural Resources, Committee on the Grand Challenges and Opportunites in Environmental Engineering for the Twenty-First Century, 2019-04-08 Environmental engineers support the well being of people and the planet in areas where the two intersect Over the decades the field has improved countless lives through innovative systems for delivering water treating waste and preventing and remediating pollution in air water and soil These achievements are a testament to the multidisciplinary pragmatic systems oriented approach that characterizes environmental engineering Environmental Engineering for the 21st Century Addressing Grand Challenges outlines the crucial role for environmental engineers in this period of dramatic growth and change The report identifies five pressing challenges of the 21st century that environmental engineers are uniquely poised to help advance sustainably supply food water and energy curb climate change and adapt to its impacts design a future without pollution and waste create efficient healthy resilient cities and foster informed decisions and actions Eco-Friendly Fiber Reinforced Polymer Composite Materials Sanjay Mavinkere Rangappa, Suchart

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