

Properties of Water Word Search

Name: _____

Period: _____

1. Water having partial positive and negative charges due to unequal sharing of electrons is known as: _____
 E R O E A F K W V H B S F M J X Q M A H
 N U O H Q C C Z G Q E J C A M L R E D I
 I S X H O F G Y B C M Q G Y U K L L F G
2. The (-) oxygen of one water molecule attracts the (+) hydrogen of another water molecule allowing them to connect to form a: _____
 L L Q Z E E T W F O Y F A V A T T Y E H
 A J F D M J S W H L F S A I J T R X L S
 K H Y O R C G E M B O M O S E C T E W P
3. Oxygen has a _____ charge
 L S L A Y L M C H A T A A F O V O Z C E
 A W L C H F L W L D V G T H U I E W P C
4. Hydrogen has a _____ charge
 L S U V M X E R D U A F E S C P Q Z O E
 F O K R E Y V Q F X V S N A G X Y E S F
5. _____ is when water molecules of all together because of hydrogen bonding and forms surface tension, the molecules typically form a droplet.
 E F K D O A J G E E I Y R H N E F W I I
 O E N S E F Y J R O P H I L E C H Y T C
 I W B U N C L S M E T E C A F B U S I H
6. A special form of cohesion that causes the surface of water to resist rupture and allows bugs to walk across it known as: _____
 C A P I L L A R Y A C T I O N G U P V E
 F D F K C T E I T H F M I O K E K O F A
 M J W C E X E J Q V L T T O V R B C E T
 B U F L F K O S T C V Y U F U Y X G C I
7. _____ is known as when molecules of all are something other than water due to all molecules all changes. (Example: bugs the droplets of water stick to the plant)
 F Z E Z A Y R X V B T Q W L O X G W V A
 E B R P E L J J C I B O H P O B O Y H X
 S H P B A W B A M Z H R V J Y S U Q U F
8. Another name for "how much plants that it reacts to water" is: _____
9. _____ is known as the tendency of water to move along the surface of a substance due to cohesion. Water molecules "pull" each other up.
10. Frozen water (ice) floats _____ because it is less dense as a solid.
11. _____ takes a lot of energy to raise the temperature of a certain amount of water.
12. Water is known to be a _____ solvent because it can dissolve many polar and ionic substances because of the partial charges within the water molecule.
13. _____ usually liquid substances equal to all dissolving or dispersing one or more other substances.
14. _____ is the water component in a solution, dissolved in the solvent.
15. Salt and sugar are dissolved in water (and in living), this means they are _____.
16. Different dissolved in water (and in living), this means it is _____.
17. _____ is typically a higher number on the pH scale, also known as "basic".
18. _____ is typically a higher P on the pH scale, not acidic, but not basic.
19. _____ This is typically a lower number on the pH scale.
20. Water is less _____ as it is cold. This is why ice floats.

Answer Key Properties Of Element

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