Mole Review Problems

Name Key Block

Solve the following problems and SHOW YOUR WORK

1. How many moles of ATOMS are there in 1.20 x 1025 methane (CH4) molecules?

1,20 × 10 molecular (1 mol 602 x 10 23 molecular) (5 externs) = 99.7 mole of externs

Calculate the molarity of a solution formed by dissolving 32.5 g of NaCl in sufficient 2. water to make 2.5 L of solution.

FU Naci = 23.0 + 35.5 + 58.5 76.1

M = " + 0.556 and + 0.22 M

How many ATOMS are contained in 32.0 grams of gaseous oxygen? 3.

FW O: 2(16.0) = 32.09/ml. 32.09 = 32.09/cml = 1.00 mol Oz 1.00 mol $O_2 \times 6.02 - 10^{25}$ micros $\times \frac{2.0 \text{ atm}}{1.00 \text{ mol}} = \frac{1.20 \times 10^{24}}{0.500}$ character of 31.75 grams. Identify the element.

4.

FU = 31.75 9 = 63.5 3/mol The element is copper.

5. Calculate the number of moles of neon gas in 54 litres at STP.

54 L = 22.4 L/mol = 2.4 mal

A sample of carbon reacts with a 9.2 gram sample of hydrogen to form a gaseous compound that has a mass of 120.0 grams. What is the molecular formula of this hydrocarbon, which has a molar mass of 26.0 grams?

9.25 = 1.09/ml - 9.2 male H Male Rose C: H := 111 (120.09 - 9.29) = 12.05/-1 = 9.2 male C C Anolomber and is 26.0 . Empired formula is CH - motor our 12:0+1.0= 13.00 so formula is C2 H

What is the mass of 2.11 x 1024 molecules of chlorine gas? 7.

2.11 × 10 24 molecules (Inol) = 3.50 mols CIE Fu Ch = 2 (35.5) = 71/2 = 710/201 3.50 mols × 71.0 5/mol = 249 g

Answer Key Pearson Chemistry Packet 15

Singhal Atul

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The Pearson Guide to Objective Chemistry for the AIEEE Singhal Atul,2009 To clear the All India Engineering Entrance Examination AIEEE students need to have a solid conceptual framework as well as adequate experience in solving original exam like questions The Pearson Guide to Objective Chemistry for the AIEEE seeks to serve this purpose by striking a unique balance between theory and practice Features such as Facts to Remember Important Guidelines Tools and Summary furnish the theoretical basis whereas practice questions arranged in levels sharpen the student's problem solving skills Designed and chiseled specifically for the AIEEE this book is the most focused manual for aspirants available

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