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1. In a cathode ray tube
A) electrons pass from the anode to the cathode.
B) electrons pass from the cathode to the anode.
C) protons pass from the anode to the cathode.
D) protons pass from the cathode to the anode.
Ans: B Category: Medium Section: 2.2
2. The scientist who determined the magnitude of the electric charge of the electron was
A) John Dalton
B) Robert Millikan
C) J. J. Thomson
D) Henry Moseley
E) R. Chang
Ans: B Category: Easy Section: 2.2
3. When J. J. Thomson discovered the electron, what physical property of the electron did he measure?
A) its charge, e
B) its charge-to-mass ratio, e/m
C) its temperature, T
D) its mass, m
E) its atomic number, Z
Ans: B Category: Easy Section: 2.2
4. Which of the following scientists developed the nuclear model of the atom?
A) John Dalton
B) Robert Millikan
C) J. J. Thomson
D) Henry Moseley
E) Ernest Rutherford
Ans: E Category: Easy Section: 2.2
5. Rutherford's experiment with alpha particle scattering by gold foil established that
A) protons are not evenly distributed throughout an atom.
B) electrons have a negative charge.
C) electrons have a positive charge.
D) atoms are made of protons, neutrons, and electrons.
E) protons are 1840 times heavier than electrons.
Ans: A Category: Medium Section: 2.2
6. Atoms of the same element with different mass numbers are called
A) ions. B) neutrons. C) allotropes. D) chemical families. E) isotopes.
Ans: E Category: Easy Section: 2.3

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Joseph Needham



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