
AQA GCSE CHEMISTRY REVISION- C1.4 CRUDE OIL AND FUELS QUESTIONS WITH CORRECT ANSWERS 2025

Hydrocarbon - CORRECT ANSWER -A compound containing only hydrogen and carbon

Alkane - CORRECT ANSWER -A saturated hydrocarbon.

Saturated Hydrocarbon - CORRECT ANSWER -There is a single carbon-carbon bond and no more hydrogen atoms can be added.

Where did crude oil come from? - CORRECT ANSWER -

Animals and plants sink to the bottom of the sea when they die, but do not decay as there is no oxygen for the bacteria to feed on. They are kept under high pressure and high temperatures so become crude oil. This process takes millions of years.

Explain Fractional Distillation - CORRECT ANSWER -

Hot crude oil vapour is pumped into the bottom (the hottest part) of the fractionating column. As crude oil is a mixture of hydrocarbons, as the vapour rises, and the temperature decreases different hydrocarbons with high-low boiling points condense and are pumped out of the column.

Properties of Hydrocarbons - CORRECT ANSWER -Long chain have:

high boiling points

low flammability

high viscosity

low volatility

a smoky flame

Short chain have:

low boiling points

high flammability

low viscosity

high volatility

Answers C1 May 21 2012 Chemistry

Arnold Weissberger



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