- 13. (4 pts) What is an ionic compound?
  - a. A compound that will dissolve in water
  - b. Two nonmetal atoms bonded together
  - c. Two metal atoms bonded together
  - d. A metal and a nonmetal atom bonded together
  - e. None of the above
- 14. (4 pts) What type of compound uses the naming system consisting of prefixes like mono-, di-, and tri-?
  - a. Ionic
  - b. Covalent
  - c. Both a and b
  - d. Neither a nor b
- 15. (4 pts) What type of compound uses a naming system that has -ide as the ending on the nonmetals?
  - a. Ionic
  - b. Covalent
  - c. Both a and b
  - d. Neither a nor b
- 16. (4 pts) How many total atoms are in this molecule: C<sub>6</sub>H<sub>12</sub>OH?
  - a. 3
  - b. 4
  - c. 19
  - d. 20
  - e. None of the above
- 17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is C<sub>1</sub>H<sub>10</sub>Cl<sub>2</sub>, what must x be?
  - a. 1
  - b. 2
  - c. 3
  - d. 4
  - e. None of the above
- (4 pts) Which of the following compounds would conduct electricity when dissolved in water? C<sub>3</sub>H<sub>6</sub>ClF C<sub>6</sub>H<sub>5</sub>CH<sub>5</sub> CS<sub>2</sub> BaSO<sub>4</sub>
  - a. C<sub>3</sub>H<sub>6</sub>CIF
  - b. C<sub>6</sub>H<sub>3</sub>CH<sub>3</sub>
  - e. CSa
  - d. BaSO<sub>4</sub>
  - e. CS2 and BaSO4

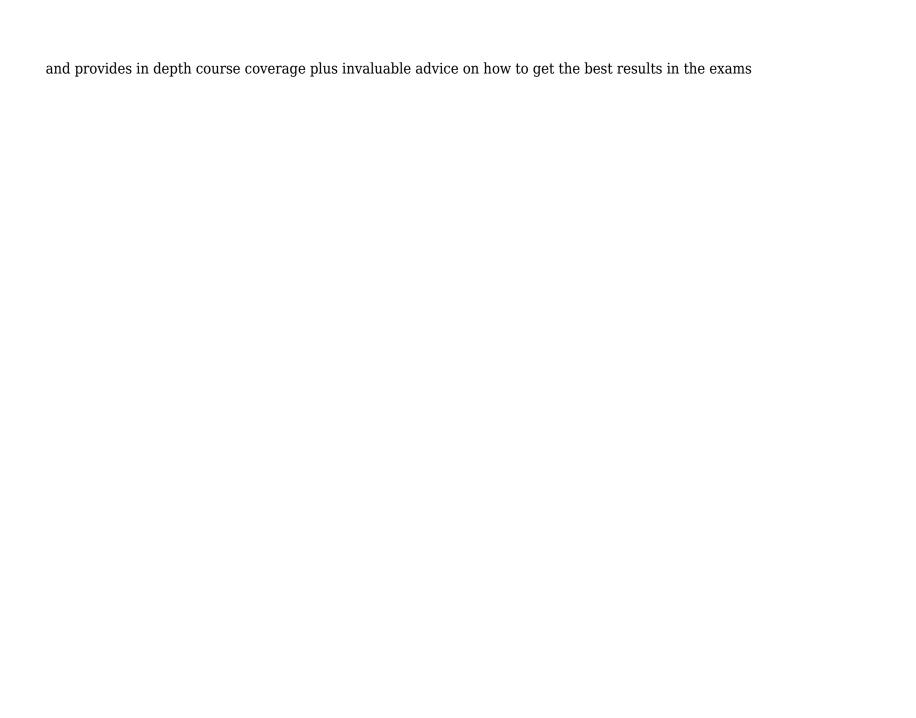
# **Answers For Chemistry Test Module 13**

**Georgios Tsaparlis** 

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