

Periodic Trends Worksheet

Directions: Use your notes to answer the following questions.

1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Oxygen < Carbon < Aluminum < Potassium

2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Neon < Aluminum < Sulfur < Oxygen

3. Why does fluorine have a higher ionization energy than iodine?

Fluorine has nine protons attracting 9 electrons which are much closer to the nucleus compared to iodine. This results in smaller size of Fluorine than iodine making it difficult to remove an electron from the atom.

4. Why do elements in the same family generally have similar properties?

Because they have same number of electrons in the outer shell (valence electrons) which take part in chemical reaction.

5. Indicate whether the following properties increase or decrease from left to right across the periodic table.

- | | |
|--|------------------|
| a. atomic radius (excluding noble gases) | Decreases |
| b. first ionization energy | Increases |
| c. electronegativity | Increases |

6. What trend in atomic radius occurs down a group on the periodic table? What causes this trend?

Atomic radius **increases** down the group on the periodic table. As we compare the elements down the group, the effective nuclear charge increases, but at the same time the outermost electrons are found in the shell that is farther away from the nucleus. Also the number of screening electrons increases. This results in reduced attraction between the nucleus and the outermost electrons.

7. What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ionization energy **increases** from left to right across the period. The atomic size becomes smaller from left to right. So it becomes harder to remove electron from the atom. Hence the energy required to do so (ionization Energy) increases.

Answer Key For Periodic Table Trend Packet

**National Institute of Standards and
Technology (U.S.)**



Answer Key For Periodic Table Trend Packet:

TRAC: Trends in Analytical Chemistry P. W. J. M. Boumans, C. J. W. Brooks, C. Burgess, 2013-09-17 Trends in Analytical Chemistry Volume 4 focuses on the advancements of processes technologies operations automation and applications of analytical chemistry The selection first offers information on bullet proof input with an IBM PC including trends in pesticide residue analysis environmental analysis using gas chromatography and ion induced spectroscopic methods for the analysis of surfaces interfaces and thin films The text then elaborates on the PROLOG an artificial intelligence language and the consideration of FORTH as a good programming environment for laboratory automation Discussions focus on computer aided optimization in high performance liquid chromatography porphyrin analysis by reversed phase high performance liquid chromatography and liquid chromatography with on line electron capture detection The text elaborates on linear regression and simple statistics calculated with the spread sheet program SUPERLAC Discussions focus on lipid membrane technology for chemical and biosensor development use of chemometrics in apportionment of air pollution sources and analytical applications of direct chromatographic enantioseparation The publication is a dependable reference for readers interested in the trends in analytical chemistry

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Publications United States. National Bureau of Standards, 1979 Publications of the National Institute of Standards and Technology ... Catalog National Institute of Standards and Technology (U.S.), 1979 The 9th Annual International Seminar on Trends in Science and Science Education (AISTSSE) 2022 ,2023-10-04 This is the ninth time we are hosting this seminar and we are proud to inform you that this seminar is an annual event in our calendar and has been held every year since 2014 This year for the third year we are holding it via Zoom meeting online meeting due to Covid 19 pandemic We are inviting internationally recognized speakers from several countries to share their latest discoveries in the fields of Biology Chemistry Physics Mathematics and Science Education Well known researchers in science and science education will share their experiences and knowledge so that we can stay up to date with the latest information This is one of the goals of this seminar As science researchers we realize the importance of information exchange among us The new information enlightens our minds and gives us ideas on what to do next in our research and how to do it This new information often becomes the foundation for our next project in particular and sets the research trends for the upcoming year in general Information exchange also keeps us updated allowing us to give and receive suggestions and critiques that will lead to better results Therefore we need a forum where we can share and exchange information Seminars conferences and other scientific gatherings are the media through which we can do this Organizer Faculty of Mathematics and Natural Sciences of Universitas Negeri Medan Where Web Seminar via Zoom Meeting When Tuesday 8th November 2022 Theme The development of industrial based research in science and science education to improve research innovation strategy Topics AISTSSE 2020 included following topics 1 Mathematics Science 2 Mathematics Education 3 Physics Science 4 Physics Education 5 Biology Science 6 Biology Education 7 Chemistry Science 8 Chemistry Education 9 Computer Science 10 Science Education Scientific Committee 1 Prof Dr Syawal Gultom M Pd Universitas Negeri Medan Indonesia 2 Prof Dr Marleen Kamperman University of Groningen Netherland 3 Prof Manihar Situmorang M Sc Ph D Universitas Negeri Medan Indonesia 4 Prof Tsunenori Mine School of Engineering Department of Electrical Engineering and Computer Science Kyushu

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