Name

# Balancing Equations Worksheet – Answers

Note to students: It is acceptable to leave spaces blank when balancing equations – blank spaces are interpreted as containing the number "1".

- Na<sub>3</sub>PO<sub>4</sub> + 3 KOH → 3 NaOH + K<sub>3</sub>PO<sub>4</sub>
- 2) MgF₂ + Li₂CO₃ → MgCO₃ + 2 LiF
- 3) P<sub>4</sub> + 3 O<sub>2</sub> → 2 P<sub>2</sub>O<sub>3</sub>
- 4) 2 RbNO<sub>3</sub> + BeF<sub>2</sub> → Be(NO<sub>3</sub>)<sub>2</sub> + 2 RbF
- 5) 2 AgNO<sub>3</sub> + Cu → Cu(NO<sub>3</sub>)<sub>2</sub> + 2 Ag
- 6) CF<sub>4</sub> + 2 Br<sub>2</sub> → CBr<sub>4</sub> + 2 F<sub>2</sub>
- 7) 2 HCN + CuSO<sub>4</sub> → H<sub>2</sub>SO<sub>4</sub> + Cu(CN)<sub>2</sub>
- 8) GaF<sub>3</sub> + 3 Cs → 3 CsF + Ga
- BaS + PtF<sub>2</sub> → BaF<sub>2</sub> + PtS (already balanced)
- 10) N<sub>2</sub> + 3 H<sub>2</sub> → 2 NH<sub>3</sub>
- 11) 2 NaF + Br<sub>2</sub> → 2 NaBr + F<sub>2</sub>
- 12) Pb(OH)<sub>2</sub> + 2 HCl → 2 H<sub>2</sub>O + PbCl<sub>2</sub>
- 13) 2 AlBr<sub>3</sub> + 3 K<sub>2</sub>SO<sub>4</sub> → 6 KBr + Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>
- 14) CH<sub>4</sub> + 2 O<sub>2</sub> → CO<sub>2</sub> + 2 H<sub>2</sub>O
- 15) 2 Na<sub>3</sub>PO<sub>4</sub> + 3 CaCl<sub>2</sub> → 6 NaCl + Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>
- 16) 2 K + Cl₂ → 2 KCl
- 17) 2 AI + 6 HCI → 3 H₂ + 2 AICIs
- 18) N<sub>2</sub> + 3 F<sub>2</sub> → 2 NF<sub>3</sub>
- 19) SO<sub>2</sub> + 2 Li<sub>2</sub>Se → SSe<sub>2</sub> + 2 Li<sub>2</sub>O
- 20) 2 NH<sub>3</sub> + H<sub>2</sub>SO<sub>4</sub> → (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>

# **Answer Key To Balancing Equations Practice**

**Greg Sherman** 

#### **Answer Key To Balancing Equations Practice:**

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